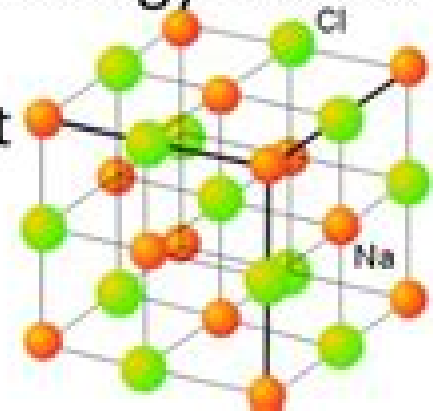


Nature of Solids

Basic Properties:

- Solids, unlike liquids, have atoms and molecules in fixed positions
 - solids have strong intermolecular forces
 - Solids have a definite shape and volume
- The density of most solids is only slightly greater than the liquid phase of the same material
 - Exception is with water
- KE of Molecules (atoms) is less than the energy it takes to separate them
 - A solid become liquid at its melting point
 - $KE = \text{intermolecular forces}$



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