

The particulate nature of matter

SOLIDS, LIQUIDS AND GASES

States of matter

Substances can exist as solids, liquids or gases.

1. Solids

In a solid the particles (molecules) vibrate about fixed positions within a close packed regular structure.

The particles cannot move in-between each other which results in a solid having a definite shape and fixed volume.

2. Liquids

When a solid is heated it may melt to form a liquid.

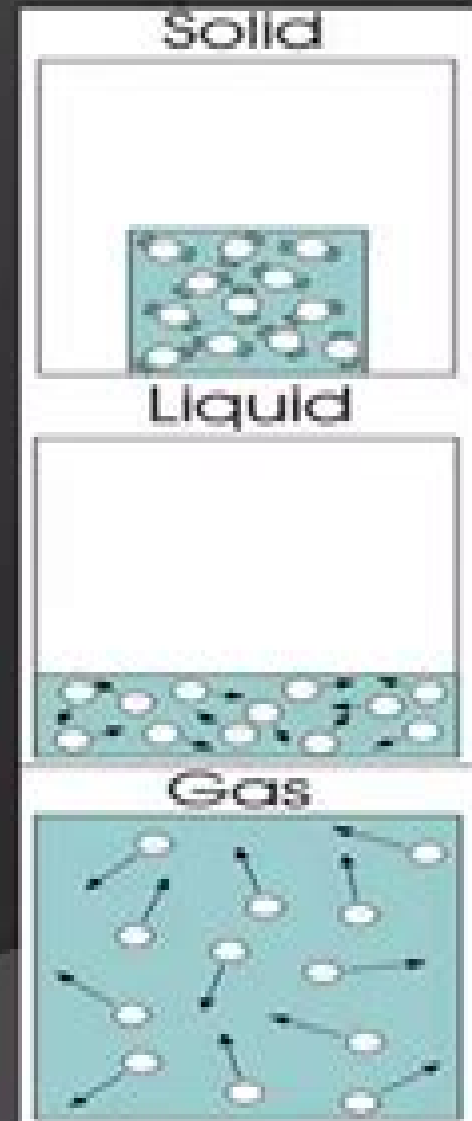
In a liquid the particles (molecules) move in-between each other. A liquid does not have a definite shape but it does have a fixed volume.

3. Gases

When a gas is heated it may evaporate or boil to form a gas.

In a gas the particles (molecules) move in-between each other and are much further apart than they are in a liquid.

A gas takes up the shape and volume of its container.



Nature Of Matter

Roger Ellman



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