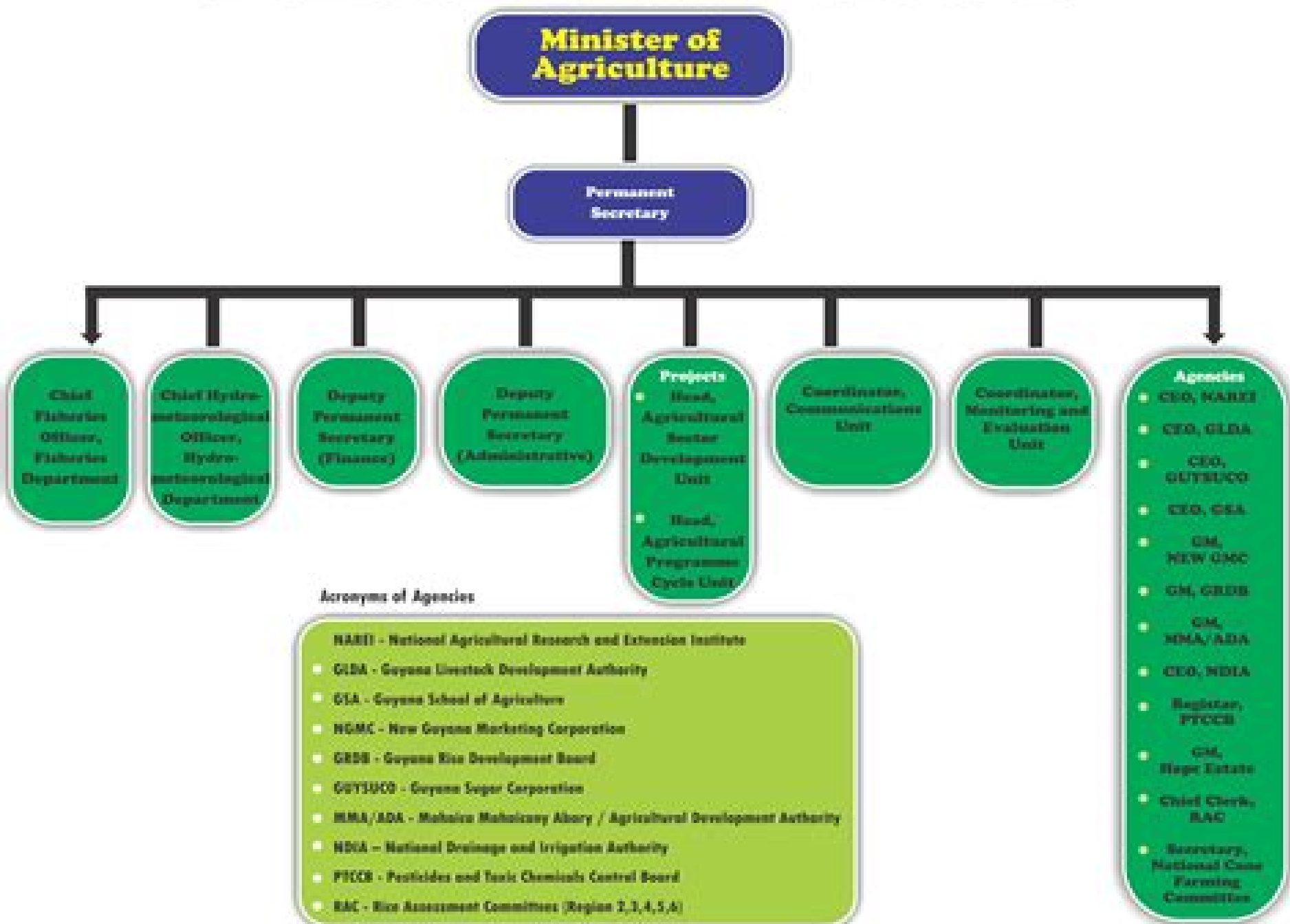


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G. Loebenstein, G. Thottappilly



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web rusting can be prevented by keeping oxygen or water away from the iron or steel oxygen can be excluded by storing the metal in an atmosphere of unreactive nitrogen or argon a desiccant is a

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web there are several cost effective ways to prevent corrosion including use non corrosive metals such as stainless steel or aluminium make sure the metal surface stays clean and dry

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web to prevent corrosion sweet systems that contain very little or no H_2S can be treated easily by using corrosion inhibitors however mitigating corrosion in systems with high levels H_2S and CO_2 is difficult this is because these systems can produce elemental sulfur and polysulfides

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web 1 barrier coatings one of the easiest and cheapest ways to prevent corrosion is to use barrier coatings like paint plastic or powder powders including epoxy nylon and urethane adhere to the metal surface to create a thin film plastic and waxes are often sprayed onto metal surfaces

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web protective coatings a corrosion study conducted by nace international now ampp shows that 50 of all corrosion costs are preventable with roughly 85 of these specific to protective coatings

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web mar 19 2021 now we will know what are the methods of preventing corrosion methods of preventing corrosion there are the following commonly used methods of preventing corrosion painting primer electroplating galvanizing chrome plating cladding silver plating gold plating nickel plating copper plating rhodium plating

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web 0 1 to the corrosive medium stop or slow down electrochemical corrosion reactions on a metal surface mechanism corrosion inhibitors work by one or more of the following mechanisms they adsorb on metal surfaces to form protective films they combine with corrosion product films to protect metal surfaces

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key aspect can be highlighted adequate access

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web jan 1 2022 introduction corrosion is a naturally occurring process which is defined as the degradation or deterioration of a substance and or its properties usually a metal over a period of time due to environmental exposure 1 this is an exergonic process as the metal tends toward the lowest possible energy state