

Quantity	Symbol	Value	Unit	Relative uncertainty (ppm)
PHYSICO-CHEMICAL CONSTANTS				
Avogadro constant	N_A, L	6.022 136 7(36)	10^{23} mol^{-1}	0.59
atomic mass constant	m_a	1.660 540 2(10)	10^{-27} kg	0.59
$m_a = \frac{1}{12} m(^{12}\text{C})$		931.494 32(28)	MeV	0.30
in electron volts: $m_a c^2 / \{e\}$	F	96 485.309(29)	C mol $^{-1}$	0.30
Faraday constant $N_A e$	$N_A h$	3.990 313 23(36)	$10^{-10} \text{ J s mol}^{-1}$	0.089
molar Planck constant	$N_A h c$	0.119 626 58(11)	J m mol $^{-1}$	0.089
molar gas constant	R	8.314 510(70)	J mol $^{-1} \text{ K}^{-1}$	8.4
Boltzmann constant R/N_A	k	1.380 658(12)	$10^{-23} \text{ J K}^{-1}$	8.5
in electron volts: $k/\{e\}$		8.617 385(73)	$10^{-5} \text{ eV K}^{-1}$	8.4
in hertz: k/h		2.083 674(18)	$10^{10} \text{ Hz K}^{-1}$	8.4
in wavenumbers: k/hc		69.503 87(59)	m $^{-1} \text{ K}^{-1}$	8.4
molar volume (ideal gas) RT/p	V_m	0.022 414 10(19)	m $^3 \text{ mol}^{-1}$	8.4
$T = 273.15 \text{ K}, p = 101\,325 \text{ Pa}$		2.686 763(23)	10^{26} m^{-3}	8.5
Loschmidt constant N_A/V_m	V_m	0.022 711 08(19)	m $^3 \text{ mol}^{-1}$	8.4
$T = 273.15 \text{ K}, p = 100 \text{ kPa}$				
Sackur-Tetrode constant	S_0/R	-1.151 693(21)		18
(absolute entropy constant) ^b		-1.164 856(21)		18
$\frac{5}{2} + \ln[(2\pi m_a k T_1/h^2)^{3/2} k T_1/p_0]$				
$T_1 = 1 \text{ K}, p_0 = 100 \text{ kPa}$				
$T_1 = 1 \text{ K}, p_0 = 101\,325 \text{ Pa}$				
Stefan-Boltzmann constant	σ	5.670 51(19)	$10^{-8} \text{ W m}^{-2} \text{ K}^{-4}$	34
$(\pi^2/60)k^4/h^3c^2$	c_1	3.741 774 9(22)	10^{-16} W m^2	0.60
first radiation constant $2\pi\hbar c^2$	c_2	0.014 387 69(12)	m K	8.4
second radiation constant hc/k				
Wien displacement law constant	b	2.897 756(24)	10^{-3} m K	8.4
$b = \lambda_{\text{max}} T = c_2/4.965\,114\,23\dots$				

Physico Chemical Constants Of Pure Volume

Carl L. Yaws



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The Yaws Handbook of Vapor Pressure Carl L. Yaws,2015-03-12 Increased to include over 25 000 organic and inorganic compounds The Yaws Handbook of Vapor Pressure Antoine Coefficients Second Edition delivers the most comprehensive and practical database source for today s petrochemical Understanding antoine coefficients for vapor pressure leads to numerous critical engineering applications such as pure components in storage vessels pressure relief valve design flammability limits at the refinery as well as environmental emissions from exposed liquids making data to efficiently calculate these daily challenges a fundamental need Written by the world s leading authority on chemical and petrochemical data The Yaws Handbook of Vapor Pressure simplifies the guesswork for the engineer and reinforces the credibility of the engineer s calculations with a single trust worthy source This data book is a must have for the engineer s library bookshelf Increase compound coverage from 8 200 to over 25 000 organic and inorganic compounds including sulfur and hydrocarbons Solve process design questions quickly from a single reliable data source Locate answers easily for multiple petrochemical related questions such as bubble point dew point temperatures and vapor liquid equilibrium *Journal of the Society of Chemical Industry* Society of Chemical Industry (Great Britain),1914 *Annotated Accession List of Data Compilations of the Office of Standard Reference Data* National Measurement Laboratory (U.S.). Office of Standard Reference Data,Herman M. Weisman,Gertrude B. Sherwood,1970

Ammonia, Amines, Phosphine, Arsine, Stibine, Silane, Germane and Stannane in Organic Solvents C. L. Young,P. G. T. Fogg,2013-10-22 This volume presents a comprehensive collection and critical evaluation of solubility data published prior to June 1983 for the compounds of the title A variety of techniques were used in the original determinations the merits of these have been considered in the evaluation of the data Emerging patterns of solubility behaviour for comparable systems are indicated The editors believe that the book will both draw attention to areas where good data are lacking and stimulate further experimental work **An Annotated Bibliography of Compiled Thermodynamic Data Sources for Biochemical and Aqueous Systems (1930 to 1975)** George Thomson Armstrong,Robert N. Goldberg,1976 101 selected references to books and journal articles Also includes some foreign language titles Alphabetical arrangement by

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Table of Contents Physico Chemical Constants Of Pure Volume

1. Understanding the eBook Physico Chemical Constants Of Pure Volume
 - The Rise of Digital Reading Physico Chemical Constants Of Pure Volume
 - Advantages of eBooks Over Traditional Books
2. Identifying Physico Chemical Constants Of Pure Volume
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physico Chemical Constants Of Pure Volume
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physico Chemical Constants Of Pure Volume
 - Personalized Recommendations
 - Physico Chemical Constants Of Pure Volume User Reviews and Ratings
 - Physico Chemical Constants Of Pure Volume and Bestseller Lists
5. Accessing Physico Chemical Constants Of Pure Volume Free and Paid eBooks
 - Physico Chemical Constants Of Pure Volume Public Domain eBooks
 - Physico Chemical Constants Of Pure Volume eBook Subscription Services

- Physico Chemical Constants Of Pure Volume Budget-Friendly Options
- 6. Navigating Physico Chemical Constants Of Pure Volume eBook Formats
 - ePub, PDF, MOBI, and More
 - Physico Chemical Constants Of Pure Volume Compatibility with Devices
 - Physico Chemical Constants Of Pure Volume Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physico Chemical Constants Of Pure Volume
 - Highlighting and Note-Taking Physico Chemical Constants Of Pure Volume
 - Interactive Elements Physico Chemical Constants Of Pure Volume
- 8. Staying Engaged with Physico Chemical Constants Of Pure Volume
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physico Chemical Constants Of Pure Volume
- 9. Balancing eBooks and Physical Books Physico Chemical Constants Of Pure Volume
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physico Chemical Constants Of Pure Volume
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Physico Chemical Constants Of Pure Volume
 - Setting Reading Goals Physico Chemical Constants Of Pure Volume
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physico Chemical Constants Of Pure Volume
 - Fact-Checking eBook Content of Physico Chemical Constants Of Pure Volume
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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