

et al. 2004; Duggal and Pasquall 2006; Fry et al. 2006; Hobbs and Fry 2006, 2007; Parra-Vega et al. 2007; Song 2006; Yang et al. 2006). However, the fundamental rheological behaviors of dispersions of most inorganic nanocylinders have yet to be characterized. Until recently, this was in part due to the limited availability of these materials. Additionally, the long lengths and high aspect ratios of these materials limit the utility of many established ensemble techniques, such as light scattering. However, recent developments in scalable synthesis methods for inorganic nanocylinders have improved availability, and advances in single-particle visualization tools have allowed the characterization of the rheological behavior of individual rods in a polydisperse sample.

In this research, the rotational and translational diffusion coefficients for dilute dispersions of germanium (Ge) nanowires in alcohols were evaluated using single-particle tracking. The measured rotational and translational diffusivities were compared to the three approaches for predicting rotational and translational diffusivity: (1) the Kirkwood equation (Doi and Edwards 1986; Kirkwood and Fleck 1956; Parra-Vega et al. 2007); (2) the Kirkwood equation using viscosity as an adjustable parameter to obtain an apparent, or best fit, viscosity (Duggal and Pasquall 2006; Li and Tang 2004); and (3) a continuum model (Duggal and Pasquall 2006; Li and Tang 2004). To the authors' knowledge, this investigation is the first time the diffusivities of individual Ge nanowires have been measured. Moreover, this is the first investigation where both the potential for wall effects and solvent viscosity effects were experimentally measured and compared to theory. Germanium nanowires were chosen because the combination of their long lengths, high aspect ratios, and rigidity are distinct from previously investigated materials including single-walled carbon nanotubes (SWNTs), *F*-actin, polyphosphine triethylsilane, microtubules, and low-aspect-ratio selenium rods (Donald and Winkle 1992; Gates et al. 1993; Maeda and Maeda 2007). In addition, germanium nanowires are a particularly important material for the development of new electronic applications; they have a lower band gap and higher electron and hole mobilities than silicon (Wang and Dai 2006).

Experimental

Germanium nanowires with hexane-treated surfaces were produced by the supersonic fluid–liquid–

solid synthesis technique (Hamrath and Koenig 2004). Rheologically dilute germanium nanowire dispersions with volume fraction $\phi = 1 \times 10^{-5}$ were prepared in methanol and 1-propanol. The solvents were obtained from Sigma Aldrich and used as received. In spite of the large density difference between germanium ($\rho_{\text{Ge}} = 5.33 \text{ g/cm}^3$) and the solvents ($\rho_{\text{Me}} = 0.792 \text{ g/cm}^3$ and $\rho_{\text{Pr}} = 0.804 \text{ g/cm}^3$), the dilute dispersions were kinetically stable for a period of several days. Transmission electron microscopy images of the as-synthesized germanium nanowires showed that the diameters ranged from 7 to 25 nm. Since dynamics are weakly dependent on the length-to-diameter ratio of the as-dispersed rods, diameter measurements were repeated for quenched dilute dispersions using atomic force microscopy (AFM). For both the methanol and 1-propanol dispersions, measurement of 100 cylindrical objects yielded diameters ranging from 7 to 60 nm, with an average diameter of 20 nm (Fig. 1). Therefore, the dispersions were predominantly individual germanium nanowires, but some bundles were also present. The similarity between the diameter distributions indicates that both solvents had the same tendency for bundle formation in the dilute concentration regime. However, at higher concentrations of $\phi = 10^{-3}$, 1-propanol was found to provide better kinetic stability.

All imaging was performed using transmitted bright-field microscopy on a Nikon 80 i optical microscope with a 60 $\times$ NA 1.40 objective and 2 $\times$ additional magnification in front of the camera. Since the smallest germanium nanowires were 2 μm in length, it was not necessary to use any fluorescent tags or a fluorescent light source. The rods were imaged using Nikon

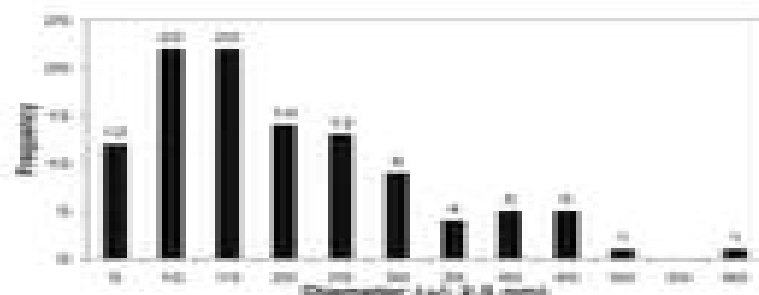


Fig. 1 Distribution of germanium nanowire diameters after dispersion in methanol obtained by AFM characterization

Nuclear Chemistry Technique Of Inor Volume 2

RJ Shavelson



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