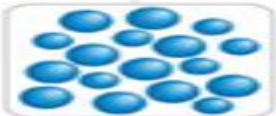


Forms of Matter and Their Properties

Solids	Liquids	Gases
In solids, tiny particles called molecules are packed closely together. They stay very close to each other. 	In liquids, the molecules are not as tightly packed as in solids. There is some space between them, which lets them move around a bit. 	In gases, the molecules are spread far apart from each other. There is a lot of empty space between them, so they can move around freely in any direction they want. 
Solids have a specific shape that doesn't change easily.	Liquids don't have a fixed or definite shape.	Gases don't have a fixed or specific shape.
They also have a specific amount of space they take up, which we call volume.	They do have a specific amount of space they occupy, which we call volume.	Gases don't have a fixed or specific volume.
Most solids are hard, meaning they are not easy to squish or bend.	Liquids can flow and move easily.	Gases can flow and move around freely.
When we move solids to a different container, they keep their shape.	When we put liquids in a container, they take the shape of that container.	Gases fill up all the available space inside a container and can be compressed easily.
Examples of solids are things like benches, books, pens, chairs and plates. 	Examples of liquids include water, ink, milk, petrol and juice. 	Examples of gases include oxygen (the gas we breathe), carbon dioxide (the gas we exhale), water vapour (the gas form of water), cooking gas (such as propane or natural gas), and nitrogen (a gas that makes up most of the air we breathe). 

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