Making Cities Work THE DYNAMICS OF URBAN INNOVATION

Edited by DAVID MORLEY STUART PROUDFOOT THOMAS BURNS



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buffers typically consist of an acid base pair with the acid and base differing by the presence or absence of a proton a conjugate acid base pair for instance one of the buffers that maintain the ph of human blood involves carbonic acid h 2 co 3 and its conjugate base the bicarbonate ion hco 3

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that is capable of donating a proton hydrogen ion to another substance a base is a molecule or ion able to accept a hydrogen ion from an acid acidic substances are usually identified by their sour taste

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