

$$H\Psi = i\hbar \frac{\partial}{\partial t} \Psi$$

Molecular Quantum Mechanics

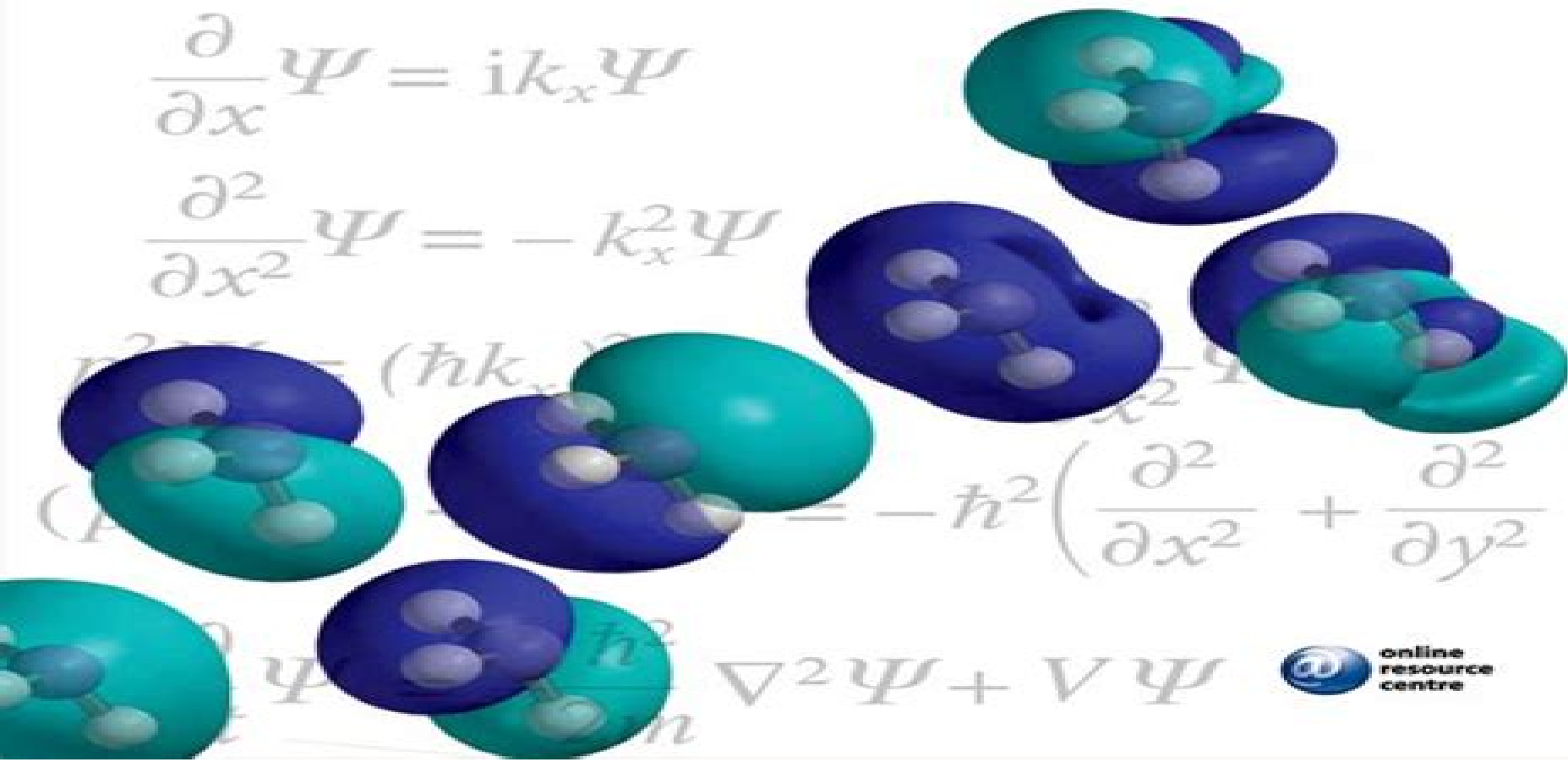
Fifth edition

Peter Atkins and Ronald Friedman

$$\frac{\partial}{\partial x} \Psi = ik_x \Psi$$

$$\frac{\partial^2}{\partial x^2} \Psi = -k_x^2 \Psi$$

$$E\Psi = -\frac{\hbar^2}{2m} \nabla^2 \Psi + V\Psi$$



Quantum Mechanics In Chemistry Edition

Daniela Niemeyer



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