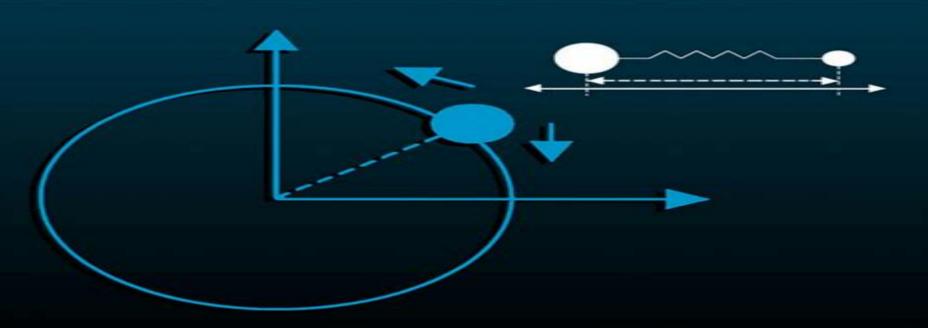
Fundamentals of Quantum Chemistry

Molecular Spectroscopy and Modern Electronic Structure Computations



Michael Mueller

Quantum Chemistry And Molecular Spectroscopy

Michael R. Mueller

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Mueller, 2001 This text is designed as a practical introduction to quantum chemistry Quantum chemistry is applied to explain and predict molecular spectroscopy and the electronic structure of atoms and molecules In addition the text provides a practical guide to using molecular mechanics and electronic structure computations including ab initio semi empirical and density functional methods The use of electronic structure computations is a timely subject as its applications in both theoretical and experimental chemical research is increasingly prevalent. This text is written in a format that fosters mastery of the subject both in competency in the mathematics and in obtaining a conceptual understanding of quantum mechanics The chemistry student's interest is maintained early on in the text where quantum mechanics is developed by applying it to molecular spectroscopy and through conceptual questions labeled as Chemical Connection Questions throughout the text labeled as Chemical Connection and Points of Further Understanding focus on conceptual understanding and consequences of quantum mechanics If an Instructor chooses these questions can be used as a basis for classroom discussion encouraging cooperative learning techniques This text provides a solid foundation from which students can readily build further knowledge of quantum chemistry in more advanced courses In cases where this is a final course in quantum chemistry this text provides the student not only with an appreciation of the importance of quantum mechanics to chemistry but also with a practical guide to using electronic structure computations Fundamentals of Quantum Chemistry Michael P. Mueller, 2015-11-11 As quantum theory enters its second century it is fitting to examine just how far it has come as a tool for the chemist Beginning with Max Planck's agonizing conclusion in 1900 that linked energy emission in discreet bundles to the resultant black body radiation curve a body of knowledge has developed with profound consequences in our ability to understand nature In the early years quantum theory was the providence of physicists and certain breeds of physical chemists While physicists honed and refined the theory and studied atoms and their component systems physical chemists began the foray into the study of larger molecular systems **Introduction to Quantum Chemistry** Clifford E. Dykstra, 1994 A brief inexpensive book on quantum chemistry and molecular spectroscopy **Computational Molecular Spectroscopy** Per Jensen, Philip R. Bunker, 2000-11-02 This book describes the use of modern computational methods in predicting high resolution molecular spectra which allows the experimental spectroscopist to interpret and assign real spectra Offers a comprehensive treatment of modern computation techniques Provides a collection of material from different areas of theoretical chemistry and physics Bridges the gap between traditional quantum chemistry and experimental molecular spectroscopy **Quantum Chemistry** Ira N. Levine, 1970

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