

Natural AOX in the river Rhine: modelling and trace analysis

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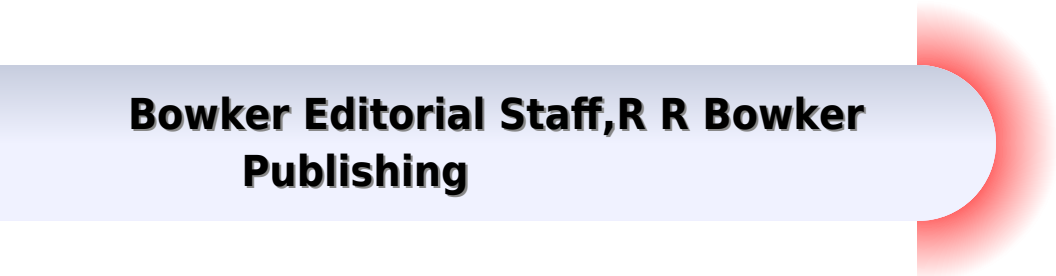
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Abstract

A new, simple model is presented to estimate natural AOX concentrations of anthropogenically polluted river systems. The model is based on the measurement of AOX/DOC ratios of relatively natural tributaries of the river system under investigation. An average natural AOX/DOC ratio of 0.35% was determined for the Rhine basin. The natural AOX concentration of a river, in a specific anthropogenically polluted part, can be calculated by multiplying the average natural AOX/DOC ratio with the DOC value at the polluted site. It appears that, in the Dutch part of the Rhine, ca. $12 \pm 4 \mu\text{g Cl/l}$ ($55 \pm 17\%$ of Rhine AOX) is of natural origin, that ca. $2 \mu\text{g Cl/l}$ (9% of Rhine AOX) is introduced by pulp mills and that ca. $8 \mu\text{g Cl/l}$ (36% of Rhine AOX) is of unknown anthropogenic origin. It is postulated that sewage works may introduce a large part of the unidentified AOX fraction. Pyrolysis-gas chromatography-mass spectrometry-single ion monitoring (Py-GC-MS-SIM) has been used to identify and quantify chlorolignosulphonic acids in the river Rhine, using the 5-chloro-2-methoxyphenol substructure. It appears that, by exception, the 5-chloro-2-methoxyphenol substructure can be identified in aquatic humic substances isolated from a single Rhine tributary.

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Naturally-Produced Organohalogens Anders Grimvall, Ed W.B. de Leer, 2012-12-06 Should the production and use of chlorine and all chlorinated organic compounds be halted in view of their adverse effects on the environment and human health Those in favour argue that certain chlorinated compounds PCBs DDT CFCs etc have large negative environmental effects The use of chlorine in disinfectants leads to the production of chloroform while bulk products PVC contribute to the production of chlorinated dibenzo p dioxins and dibenzofurans when they are burned Those against argue that chlorine and many chlorinated compounds are essential in the control of human health the prevention of disease transmitted through drinking water that has not been disinfected and that chlorinated compounds are indispensable intermediates in many production processes representing a vast economic value But such discussions often ignore the fact that Nature contributes significantly to the production of chlorinated organic compounds More than 1000 such compounds are known and their contribution to the biogeochemical cycling of chlorine is underestimated Chlorine is organically bound in large quantities to humic materials and natural production mechanisms are known for low molecular weight compounds methyl chloride chloroform chlorinated dibenzo p dioxins and dibenzofurans The role of these compounds in the environment is largely unknown *Naturally Produced Organohalogens* gives a complete overview of the present state of knowledge on the subject giving a much needed balance to the argument sketched out above

Handbook on Biodegradation and Biological Treatment of Hazardous Organic Compounds M.H. van Agteren, Sytze Keuning, D. Janssen, 2013-04-17 The introduction of synthetic organic chemicals into the environment during the last few decades has given rise to major concern about the ecotoxicological effects and ultimate fate of these compounds The pollutants that are considered to be most hazardous because of their intrinsic toxicity high exposure level or recalcitrant behavior in the environment have been placed on blacklists and other policy priority lists The fate of synthetic compounds that enter the environment is mainly determined by their rate of biodegradation which therefore also has a major effect on the degree of bioaccumulation and the risk of ecotoxicological effects The degree and rate of biodegradation is also of critical importance for the feasibility of biological techniques to clean up contaminated sites and waste streams The biodegradation of xenobiotics has thus been the subject of numerous studies which resulted in thousands of publications in scientific journals books and conference proceedings These studies led to a deeper understanding of the diversity of biodegradation processes As a result it has become possible to enhance the rate of degradation of recalcitrant pollutants during biological treatment and to design completely new treatment processes At present much work is being done to expand the range of pollutants to which biodegradation can be applied and to make treatment techniques less expensive and better applicable for waste streams which are difficult to handle

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