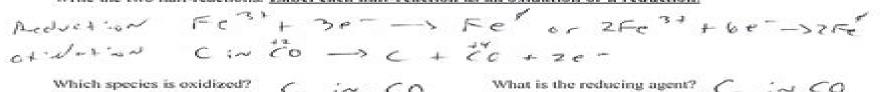


Write the two half-reactions. Label each half-reaction as an oxidation or a reduction.



 In the smelting of iron from iron oxide according to the equation Fe₂O₂(s) + 3 CO(g) → 2 Fe(s) + 3 CO₂(g), which is the oxidizing agent?

(a) C in CO (b) O in Fe₂O₃ (c) C in CO₂ (d) Fe metal (e) Fe in Fe₂O₂

- Identify and justify the Bronsted-Lowry base in the reaction: NH₃+ H-Br → NH₄* Br
 (a) NH₃ because it is a proton donor (b) HBr because it is a proton donor (c) 78H₃ because it is a proton acceptor
- 3. A solution is prepared by dissolving 1.28 g of NaNO₃ in enough water to make 500 mL of solution. What is the concentration of the solution in molarity? (The unit of molarity is mols / liter given the symbol M)

 2.2.9398

 1.2.8.3.Vanie3 | 1 mol Naple3 | 0.0151me/ = 0.551ml = 15.9534 × 3 | 54.99475 NeNG NanO3 | 0.0301 M
 - What is the weight (in grams) of NaNO₃ in a 0.500 L bottle of 2.15 M NaNO₃?

Problems For General Chemistry Qualita

M Walker

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