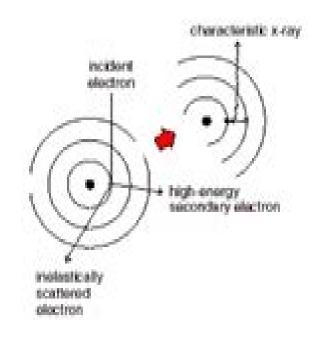
K-shell Emission Radiation

- Remember that atoms have their electrons arranged in closed "shells" of different energies. The K-shell is the lowest energy state of an atom.
- An incoming electron can give a K-shell electron enough energy to knock it out of its energy state.
- About 0.1% of the electrons produce K-shell vacancies; most produce heat. Then, a tungsten electron of higher energy (from an outer shell) can fall into the K-shell.
- The energy lost by the falling electron shows up as an emitted Xray photon. Meanwhile, higher energy electrons fall into the vacated energy state in the outer shell, and so on.
- After losing an electron, an atom remains ionized for a very short time (about 10⁻¹⁴ second) and thus an atom can be repeatedly ionized by the incident electrons which arrive about every 10⁻¹² second.
- Generally, K-shell emission produces higher-intensity X-rays than Bremsstrahlung, and the X-ray photon comes out at a single wavelength.



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