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Made Easy

Kelly Riddle

Private Investigating Made Easy How To Conduct Investigations Using Public Records And More

**United States. Congress. Joint
Committee to Investigate the Interior
Department and Forestry Service**

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web we utilize the equation that relate amount remaining initial mass and number of half lives $n = \frac{t}{T} \times \log_2 \frac{m_0}{m}$ 125 g 5 what is the half life of an isotope that is 80 remained after 16 days solution remaining 80 100 therefore $n = \frac{t}{T} \times \log_2 \frac{m_0}{m}$ now using the half life equation in example 1 we have

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