

# TABLEAU PÉRIODIQUE DES ÉLÉMENTS

| 1                       | 2                         | 3                         | 4                 | 5                   | 6                 | 7                   | 8                  | 9                   | 10                  | 11                  | 12                 | 13                       | 14                      | 15                     | 16                      | 17                       | 18                         |
|-------------------------|---------------------------|---------------------------|-------------------|---------------------|-------------------|---------------------|--------------------|---------------------|---------------------|---------------------|--------------------|--------------------------|-------------------------|------------------------|-------------------------|--------------------------|----------------------------|
| 1<br>IA<br>H<br>1,00794 | 2<br>IIA<br>He<br>4,00260 |                           |                   |                     |                   |                     |                    |                     |                     |                     |                    | 3<br>IIIA<br>B<br>10,811 | 4<br>IVA<br>C<br>12,011 | 5<br>VA<br>N<br>14,007 | 6<br>VIA<br>O<br>15,999 | 7<br>VIIA<br>F<br>18,998 | 8<br>VIIIA<br>Ne<br>20,180 |
| 3<br>Li<br>6,941        | 4<br>Be<br>9,012          |                           |                   |                     |                   |                     |                    |                     |                     |                     |                    | 5<br>Al<br>26,982        | 6<br>Si<br>28,086       | 7<br>P<br>30,974       | 8<br>S<br>32,06         | 9<br>Cl<br>35,45         | 10<br>Ar<br>39,948         |
| 4<br>K<br>39,098        | 5<br>Ca<br>40,078         | 6<br>Sc<br>44,956         | 7<br>Ti<br>47,88  | 8<br>V<br>50,942    | 9<br>Cr<br>51,996 | 10<br>Mn<br>54,938  | 11<br>Fe<br>55,845 | 12<br>Co<br>58,933  | 13<br>Ni<br>58,69   | 14<br>Cu<br>63,546  | 15<br>Zn<br>65,38  | 16<br>Ga<br>69,723       | 17<br>Ge<br>72,64       | 18<br>As<br>74,922     | 19<br>Se<br>78,96       | 20<br>Br<br>79,904       | 21<br>Kr<br>83,80          |
| 5<br>Rb<br>85,468       | 6<br>Sr<br>87,62          | 7<br>Y<br>88,906          | 8<br>Zr<br>91,224 | 9<br>Nb<br>92,906   | 10<br>Mo<br>95,94 | 11<br>Tc<br>98,906  | 12<br>Ru<br>101,07 | 13<br>Rh<br>102,91  | 14<br>Pd<br>106,42  | 15<br>Ag<br>107,87  | 16<br>Cd<br>112,41 | 17<br>In<br>114,82       | 18<br>Sn<br>118,71      | 19<br>Sb<br>121,76     | 20<br>Te<br>127,60      | 21<br>I<br>126,905       | 22<br>Xe<br>131,29         |
| 6<br>Cs<br>132,91       | 7<br>Ba<br>137,33         | 8<br>La-Lu<br>Lanthanides | 9<br>Hf<br>178,49 | 10<br>Ta<br>180,948 | 11<br>W<br>183,84 | 12<br>Re<br>186,207 | 13<br>Os<br>190,23 | 14<br>Ir<br>192,222 | 15<br>Pt<br>195,084 | 16<br>Au<br>196,967 | 17<br>Hg<br>200,59 | 18<br>Tl<br>204,384      | 19<br>Pb<br>207,2       | 20<br>Bi<br>208,98     | 21<br>Po<br>209         | 22<br>At<br>210          | 23<br>Rn<br>222            |
| 7<br>Fr<br>223          | 8<br>Ra<br>226            | 9<br>Ac-Lr<br>Actinides   | 10<br>Rf<br>261   | 11<br>Db<br>262     | 12<br>Sg<br>266   | 13<br>Bh<br>264     | 14<br>Hs<br>277    | 15<br>Mt<br>268     | 16<br>Ds<br>271     | 17<br>Rg<br>272     | 18<br>Uub<br>285   | 19<br>Uut<br>288         | 20<br>Uuq<br>291        | 21<br>Uup<br>293       | 22<br>Uuh<br>295        | 23<br>Uus<br>294         | 24<br>Uuo<br>293           |

## LANTHANIDES

|                     |                    |                     |                    |                   |                    |                     |                    |                     |                    |                     |                     |                     |                     |                     |
|---------------------|--------------------|---------------------|--------------------|-------------------|--------------------|---------------------|--------------------|---------------------|--------------------|---------------------|---------------------|---------------------|---------------------|---------------------|
| 57<br>La<br>138,905 | 58<br>Ce<br>140,12 | 59<br>Pr<br>140,908 | 60<br>Nd<br>144,24 | 61<br>Pm<br>(145) | 62<br>Sm<br>150,36 | 63<br>Eu<br>151,964 | 64<br>Gd<br>157,25 | 65<br>Tb<br>158,925 | 66<br>Dy<br>162,50 | 67<br>Ho<br>164,930 | 68<br>Er<br>167,259 | 69<br>Tm<br>168,934 | 70<br>Yb<br>173,054 | 71<br>Lu<br>174,967 |
|---------------------|--------------------|---------------------|--------------------|-------------------|--------------------|---------------------|--------------------|---------------------|--------------------|---------------------|---------------------|---------------------|---------------------|---------------------|

## ACTINIDES

|                 |                     |                     |                    |                 |                 |                 |                 |                 |                 |                 |                  |                  |                  |                  |
|-----------------|---------------------|---------------------|--------------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|------------------|------------------|------------------|------------------|
| 89<br>Ac<br>227 | 90<br>Th<br>232,038 | 91<br>Pa<br>231,036 | 92<br>U<br>238,029 | 93<br>Np<br>237 | 94<br>Pu<br>244 | 95<br>Am<br>243 | 96<br>Cm<br>247 | 97<br>Bk<br>247 | 98<br>Cf<br>251 | 99<br>Es<br>252 | 100<br>Fm<br>257 | 101<br>Md<br>258 | 102<br>No<br>259 | 103<br>Lr<br>262 |
|-----------------|---------------------|---------------------|--------------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|------------------|------------------|------------------|------------------|

(1) La masse atomique relative est donnée avec cinq chiffres significatifs. Pour les éléments qui n'ont pas de radioisotope stable, la valeur entre parenthèses indique le nombre de protons de l'isotope de l'élément ayant le plus grand nombre de protons.

Toutes les données sont données pour les éléments les plus stables connus. Les masses atomiques sont données.

(2) Les configurations électroniques pour lesquelles aucune exception n'est donnée.

ELECTRONégativité  
DES ÉLÉMENTS

POSITION DES  
Sous-niveaux

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