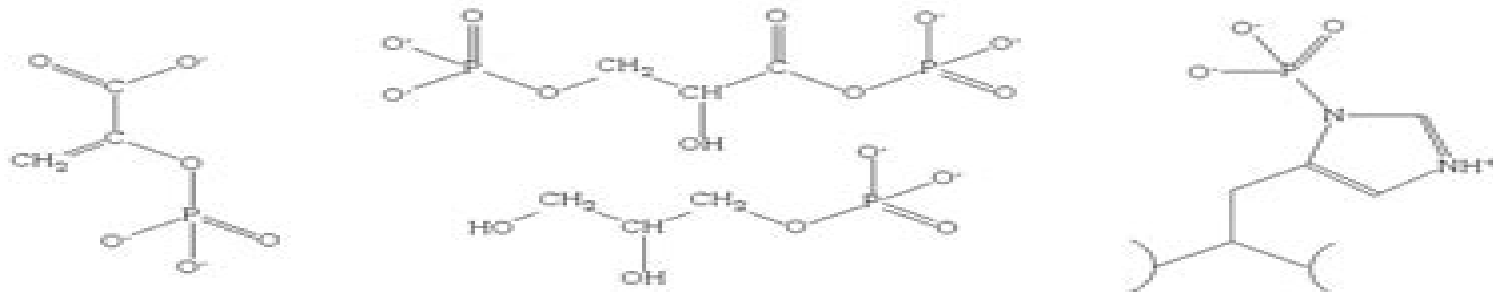


(11) Place an 'x' on each of the high energy bonds shown in the structures below.



(12) Consider the oxidation-reduction reaction between cytochrome b and ubiquinol.

(a) What is the ΔG° for the reduction of cytochrome b by ubiquinol? (Note: you will need to use the tables of standard reduction potentials)

(b) Under what conditions will the reduction of cytochrome b by ubiquinol be favourable at 298K?

(13) Indicate whether the following statements are true or false. If false, change a single word or number to make the statement true.

(a) "(6- ^{13}C)Glucose (ie. C6 label) is converted to (1- ^{13}C)Glyceraldehyde-3-phosphate during the preparatory phase of glycolysis."

(b) "Phosphoglycerate kinase requires catalytic amounts of 2,3-bisphosphoglycerate for activity."

(c) "Glyceraldehyde-3-phosphate dehydrogenase couples the addition of phosphate to glyceraldehyde-3-phosphate and the reduction of NAD^+ ."

(14) Identify the glycolytic enzyme with the following properties:

(a) Forms a "Schiff's base" intermediate during its catalytic cycle.

(b) Synthesized in an inactive form.

(c) Is inhibited by alkylating agents such as iodoacetate.

(d) Is a lyase.

(15) Answer the following questions regarding glycolysis:

(a) What is the overall chemical reaction for the 'preparatory' phase of glycolysis?

(b) What is substrate-level phosphorylation?

(c) Under cellular conditions, which glycolytic enzymes are likely to be regulatory targets?

(d) Which glycolytic intermediates are 'high-energy' compounds?

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Jane M. Magill



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