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Colonna,Antonio D'Angola,2011-12-02 Fundamental Aspects of Plasma Chemical Physics Thermodynamics develops basic and advanced concepts of plasma thermodynamics from both classical and statistical points of view After a refreshment of classical thermodynamics applied to the dissociation and ionization regimes the book invites the reader to discover the role of electronic excitation in affecting the properties of plasmas a topic often overlooked by the thermal plasma community Particular attention is devoted to the problem of the divergence of the partition function of atomic species and the state to state approach for calculating the partition function of diatomic and polyatomic molecules The limit of ideal gas approximation is also discussed by introducing Debye Huckel and virial corrections Throughout the book worked examples are given in order to clarify concepts and mathematical approaches This book is a first of a series of three books to be published by the authors on fundamental aspects of plasma chemical physics The next books will discuss transport and kinetics **Mathematical Reviews** ,2004 *Library Journal* ,1962 **The Quantum Theory of Line-spectra** Niels Bohr,1922 **School Library Journal** ,1961 Whitaker's Cumulative Book List ,1964 On the Quantum Theory of Line-Spectra, Vol. 2 Niels Bohr,2016-09-30 Excerpt from On the Quantum Theory of Line Spectra Vol 2 On the Hydrogen Spectrum In Part III the problem of the series spectra of other elements will be treated from a similar point of view As pointed out by the writer in an earlier paper a simple explanation of the pronounced analogy between these spectra and the hydrogen spectrum is offered by the fact that the atomic systems involved in the emission of the spectra under consideration in a certain sense may be regarded as a perturbed hydrogen atom On the other hand a clue to the interpretation of the characteristic difference between the hydrogen spectrum and the spectra of other elements was first obtained by sommerfeld's theory of the stationary states of central systems referred to above As shown by sommerfeld it is possible on this theory to account in general outlines for the well known laws governing the frequencies of the series spectra of the elements and as it will be shown in Part III it is also possible on the basis of the formal relation between the quantum theory and the ordinary theory of radiation in this way to obtain a simple interpretation of the laws governing the remarkable differences in the intensities with which the various series of lines appear which on the combination principle would constitute the complete spectra under consideration As regards the detailed discussion of these spectra however it is necessary to bear in mind that the part played by the inner electrons in the atoms of the elements in question forms a far more intricate problem than the perturbing effect of a fixed external field on the hydrogen atom For the treatment of this problem the theory of conditionally periodic systems based on the conditions 22 does not seem to suffice while as it will be shown in Part III it appears that the method of perturbations exposed in the following lends itself naturally also to this case About the Publisher Forgotten Books

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On the Quantum Theory of Line-Spectra Niels Henrik David Bohr, 2015-06-02 Excerpt from On the Quantum Theory of Line Spectra Vol 1 In an attempt to develop certain outlines of a theory of line spectra based on a suitable application of the fundamental ideas introduced by Planck in his theory of temperature radiation to the theory of the nucleus atom of Sir Ernest Rutherford the writer has shown that it is possible in this way to obtain a simple interpretation of some of the main laws governing the line spectra of the elements and especially to obtain a deduction of the well known Balmer formula for the hydrogen spectrum The theory in the form given allowed of a detailed discussion only in the case of periodic systems and obviously was not able to account in detail for the characteristic difference between the hydrogen spectrum and the spectra of other elements or for the characteristic effects on the hydrogen spectrum of external electric and magnetic fields Recently however a way out of this difficulty has been opened by Sommerfeld who by introducing a suitable generalization of the theory to a simple type of non periodic motions and by taking the small variation of the mass of the electron with its velocity into account obtained an explanation of the fine structure of the hydrogen lines which was found to be in brilliant conformity with the measurements Already in his first paper on this subject Sommerfeld pointed out that his theory evidently offered a clue to the interpretation of the more intricate structure of the spectra of other elements Briefly afterwards Epstein and Schwarzschild independent of each other by adapting Sommerfeld's ideas to the treatment of a more extended class of non periodic systems obtained a detailed explanation of the characteristic effect of an electric field on the hydrogen spectrum discovered by Stark About the Publisher Forgotten Books publishes hundreds of thousands of rare and classic books Find more at www.forgottenbooks.com This book is a reproduction of an important historical work Forgotten Books uses state of the art technology to digitally reconstruct the work preserving the original format whilst repairing imperfections present in the aged copy In rare cases an imperfection in the original such as a blemish or missing page may be replicated in our edition We do however repair the vast majority of imperfections successfully any imperfections that remain are intentionally left to preserve the state of such historical works

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