

Chemistry

Topic: Significant Figures(SF)

What are Significant Figures??

- Significant Figures are the result in a measurement being more accurate and it also refers to the digits being measured.
- Significant Figures are useful in indicating the exactness of a measurement: in doing so numbers stay the same regardless of how many times the measurement has been taken.
- Significant Figures have an "Uncertain Digit" aka "the last number," which constantly changes upon doing measurements. It is the only number that changes in Significant Figures.

Example: Measurement 1 – SF = 1.23**9**

Measurement 1 – SF = 1.23**7**

Measurement 3 - SF = 1.23**8**

Accuracy:

- ☐ Refers to the closeness of a measured value to the correct value.

Precision:

- ☐ Refers to the closeness of two or more measurements to one another.

"NOTE!!! When working with SF, there are certain numbers that can and can not be considered a SF."

Numbers that ARE considered Significant Figures(SF) are:

- Numbers the are **NOT** zero = 1, 2, 3, 4, 5.....etc.
- Numbers with zero in **between** = 30005 or 20000007.

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Velavancorresponding



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