Reactivity of Solids

Solid state chemistry deals with the study of the synthesis, structure and physical properties of solid materials and continues to play an astounding role in many disciplines.

Solid state reactions may involve one or more of the following elementary (rate-limiting) steps:

- (i) Sorption phenomena (adsorption and desorption) of gaseous species on the solid surface;
- (ii) Chemical reaction on an atomic scale (homogeneous or interphase reactions);
- (iii) Nucleation of a new phase (in the bulk or at the surface of the reacting solid);
- (iv) Transport phenomena through solids (diffusion and migration).

Solid state reactions may be classified into four categories.

(a) Reactions of single solids

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Philip Charles Harry Mitchell, A. Seaman

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Proceedings of the Fifth International Symposium on Solid Oxide Fuel Cells (SOFC-V) U. Stimming,1997 Nuclear Science Abstracts ,1971-03 Kinetics of Reactions in Ionic Systems T. J. Gray, V. D. Fréchette,2013-12-14 Index to Conferences Relating to Nuclear Science Willie E. Clark, Doris H. McGinnis, 1968 Russian Chemical Reviews ,2006

Magnetism and Metallurgy of Soft Magnetic Materials Chih-Wen Chen, 2013-02-19 DIVDetailed theoretical study and a practical survey for solid state physicists engineers graduate students Ferromagnetism and ferrimagnetism magnetization and domain structure much more 227 figures div Sci-tech Book Profiles, 1965 Includes title page table of contents list of contributors preface and all indexes of each book Chemical Abstracts Service Source Index American Chemical Society. Chemical Abstracts Service, 1907 A key source to journal and conference abbreviations in the sciences Although it focuses on chemistry other scientific and engineering disciplines are also well represented In addition to the abbreviation and full title each entry also contains publishing info title changes language and frequency of publication and libraries owning that title Over 130 000 entries representing more than 70 000 publications dating back to 1907 are included

<u>Isoconversional Kinetics of Thermally Stimulated Processes</u> Sergey Vyazovkin, 2015-03-18 The use of isoconversional kinetic methods for analysis of thermogravimetric and calorimetric data on thermally stimulated processes is quickly growing in popularity The purpose of this book is to create the first comprehensive resource on the theory and applications of isoconversional methodology The book introduces the reader to the kinetics of physical and chemical condensed phase processes that occur as a result of changing temperature and discusses how isoconversional analysis can provide important kinetic insights into them The book will help the readers to develop a better understanding of the methodology and promote its efficient usage and successful development Chemistry & Uses of Molybdenum Philip Charles Harry Mitchell, A. Kristallzüchtung Klaus Thomas Wilke, 1973 Treatise on Analytical Chemistry Izaak Maurits Seaman, 1977 Kolthoff, Philip Juliber Elving, 1993 Dictionary Catalog of the Research Libraries of the New York Public Library, **1911-1971** New York Public Library. Research Libraries,1979 Treatise on Analytical Chemistry: Theory and practice. v Izaak Maurits Kolthoff, Philip J. Elving, 1959 AFOSR Chemical & Atmospheric Sciences Program Review United States. Air Force. Directorate of Chemical and Atmospheric Sciences, **Chemistry and Kinetics of the Sulphating Roasting of Uranium-bearing Silicates** G. Asmund, 1971 Future Energy Conferences and Symposia, 1989 Developments in *Inorganic Nitrogen Chemistry* Charles B. Colburn,1966 *Proceedings of the Fifth International Conference on Erosion by* Liquid and Solid Impact, 1979 Laser-Induced Chemical Processes Jeffrey I. Steinfeld, 2012-12-06 The possibility of initiating chemical reactions by high intensity laser excitation has captured the imagination of chemists and physicists as well as of industrial scientists and the scientifically informed public in general ever since the laser first became available Initially great hopes were held that laser induced chemistry would revolutionize synthetic chemistry making possible bond

specific or mode specific reactions that were impos sible to achieve under thermal equilibrium conditions Indeed some of the early work in this area typically employing high power continuous wave sources was interpreted in just this way With further investigation however a more conservative picture has emerged with the laser taking its place as one of a number of available methods for initiation of high energy chemical transformations Unlike a number of these methods such as flash photolysis shock tubes and electron beam radiolysis the laser is capable of a high degree of spatial and molecular localization of deposited energy which in turn is reflected in such applications as isotope enrichment or localized surface treatments The use of lasers to initiate chemical processes has led to the discovery of several distinctly new molecular phenomena foremost among which is that of multiple photon excitation and dissociation of polyatomic molecules This research area has received the greatest attention thus far and forms the focus of the present volume

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