4. Use the following data to calculate the reaction rate law for the system

$$NO_{(g)} + H_{2(g)} \longrightarrow HNO_{2(g)}$$

a. Find the order of the reactant with respect to NO and H2 (2 marks)

b. Calculate the value of rate constant. (1 mark)

c. Write the rate law (1 mark)

Experiment	NO (mol/L)	H2 (mol/L)	Initial Rate of Reaction (mol/(L·s))
1	0.001	0.004	0.002
2	0.002	0.004	0.008
3	0.003	0.004	0.018
4	0.004	0.001	0.008
5	0.004	0.002	0.016
6	0.004	0.003	0.024

Reaction Rate Compilation For The H O N System

Michal Rosen-Zvi

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