

READERS' GUIDE
to periodical literature
(UNABRIDGED)

MARCH 1969 TO FEBRUARY 1970

Readers Guide To Periodical Literature 1989

AW Chickering



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web we utilize the equation that relate amount remaining initial mass and number of half lives $n = \frac{t}{T} \times \log_2 \frac{A_0}{A}$

125 g 5 what is the half life of an isotope that is 80 remained after 16 days solution remaining 80 100 therefore $n = \frac{t}{T} \times \log_2 \frac{A_0}{A}$

now using the half life equation in example 1 we have

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