

a) Reverse Osmosis (RO)

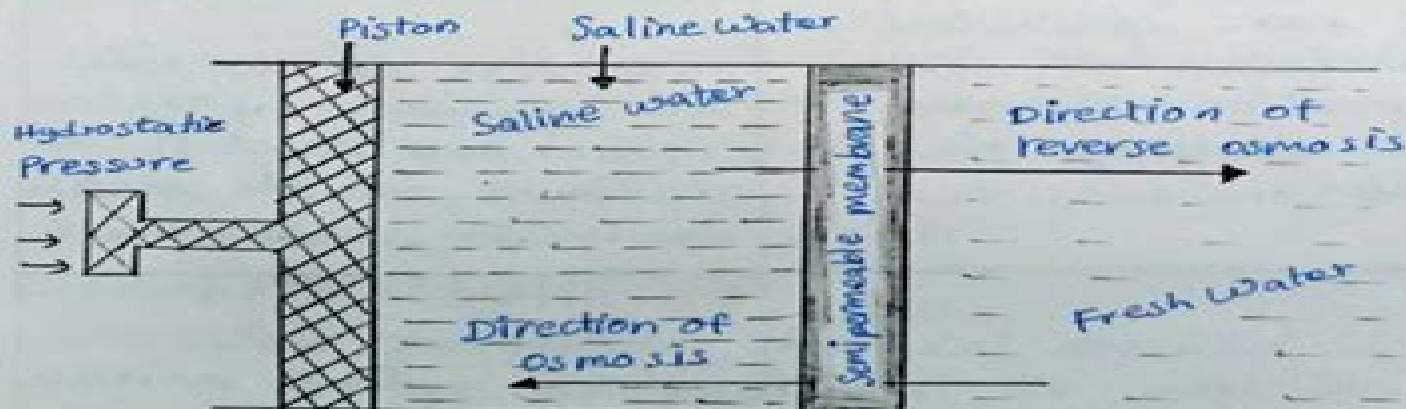
Osmosis - Definition

When two solutions of different concentrations are separated by a semi-permeable membrane, solvent (water) flows from a region of lower concentration to higher concentration. This process is called osmosis. The driving force in this phenomenon is called osmotic pressure.

Reverse Osmosis - Definition

If a hydrostatic pressure in excess of osmotic pressure is applied on the higher concentration side, the solvent flow is reversed i.e., solvent flows from higher concentration to lower concentration. This process is called reverse osmosis.

Thus, in the process of reverse osmosis pure water is separated from salt water. This process is also known as super-filtration. The membranes used are cellulose acetate, cellulose butyrate, etc.



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