

4. Use the following data to calculate the reaction rate law for the system



- Find the order of the reactant with respect to NO and H₂ (2 marks)
- Calculate the value of rate constant. (1 mark)
- Write the rate law (1 mark)

Experiment	NO (mol/L)	H ₂ (mol/L)	Initial Rate of Reaction (mol/(L·s))
1	0.001	0.004	0.002
2	0.002	0.004	0.008
3	0.003	0.004	0.018
4	0.004	0.001	0.008
5	0.004	0.002	0.016
6	0.004	0.003	0.024

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