3. В сосуд, содержащий воду массой  $m_1 = 200$  г при температуре  $t^\circ_1 = 20^\circ$ С, опустили медное тело массой  $m_2 = 100$  г, имеющее температуру  $t^\circ_2 = 120^\circ$ С. Какая общая температура установится в сосуде? Нагреванием сосуда пренебречь. Удельные теплоёмкости воды и меди равны соответственно  $c_1 = 4200 \ Дж/(кг\cdot 1^\circ C), c_2 = 380 \ Дж/(кг\cdot 1^\circ C).$ 

Решение. Определим направление теплообмена в системе "холодная вода + горячее медное тело": медное тело отдаёт тепло; холодная вода получает тепло.

 $Q_1 = m_1 c_1 (t^\circ - t_1^\circ) -$  количество теплоты, полученное холодной водой.

 $Q_2 = m_2 c_2 (t_2^\circ - t^\circ) -$  количество теплоты, отданное горячим медным телом.

В соответствии с уравнением теплового баланса (всеми потерями энергии пренебрегаем): Q<sub>1</sub> = Q<sub>2</sub>.

Раскроем полученное равенство:

$$m_1c_1(t^{\circ}-t_1^{\circ})=m_2c_2(t_2^{\circ}-t^{\circ}).$$

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**C** Cleary

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Decoding Raschet Parametrov Termogidrodinamicheskogo Vzaimodeistviia Kharakteristik Teplovogo Balansa I Temperatury Vody V Nestratifitsirovannykh Vodoemakh: Revealing the Captivating Potential of Verbal Expression

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