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point is to establish the relative role of nonclassical and classical ions in various chemical processes. The author has followed P Bartlett's advice 1 that when setting forth the achievements of the human mind one should see how we came to the modern understanding of a given problem how we know what we know. The theory of nonclassical ions offers an explanation of many unique chemical stereochemical and kinetic peculiarities of bicyclic compounds. It has expanded our knowledge on chemical bonds in carbocations by introducing electron deficient bonds as in boron hydrides. It has accounted for many rearrangements of stable cations. As a side result our knowledge has been extended about ionization processes in a solution as well as about stereochemical methods 2 Main Terms of Nonclassical Carbocations. In 1939 Hevell Salas and Wilson 23 assumed an intermediate bridge ion 2 to be formed when camphene hydrochloride 1 is rearranged into isobornyl chloride 3. This happened 17 years after Meerwein first postulated the intermediate formation of carbonium ions in chemical reactions.

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