

Chemistry 122 (Matter and Energy) Spring 2023
Discussion Worksheet Week 6 (Feb 13-17)

Chapter 4(3-5) Learning Outcomes:

Acid-Base Reactions

- Identify acids and bases in acid-base reactions based on H^+ transfer.
- Distinguish strong vs weak acids and bases.
- Write balanced molecular, total ionic, and net ionic equations for acid-base reactions.
- Perform limiting reactant stoichiometry calculations with acid-base reactions such as in titrations.

Redox Reactions

- Assign oxidation numbers to elements in ionic and covalent compounds.
- Identify redox reactions from changes of oxidation state.
- In redox reactions, use oxidation states to identify what is being oxidized, what is being reduced, and how many electrons are being transferred.
- Identify oxidizing and reducing agents in redox reactions.
- Write molecular, total ionic, and net ionic equations for redox reactions when products given.
- Use activity series to predict reactions of metals with acids and halogens with halides.

Textbook Chapter Readings in Silberberg, 9e: Chap 4(3-5)

Student Directions: Complete your Chapter Readings in textbook listed above and review any related lecture notes. Then write out complete solutions neatly on separate paper or notebook with your *best effort* and bring your work to discussion along with any specific questions where you had difficulties.

Best to save your work as part of preparing for future exams.

-H-
-N-
-O-
-Ba-

1. Write balanced molecular, complete ionic and net ionic balanced equations for the following acid-base, or gas-forming reactions. Include states of matter.

a) mixing aqueous solutions of nitric acid and barium hydroxide



b) mixing aqueous solutions of acetic acid and potassium hydroxide



c) adding aqueous hydrobromic acid to solid sodium carbonate



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