

7. In the addition reaction of aldehydes and ketones, sodium bisulfite was used to detect the presence of aldehydes and ketones. Describe what happened when sodium bisulfite was added to acetone in the addition reaction? **White precipitate was formed.**

1 The reactions of the enolate anions are also known as the haloform reactions. This is a test for methyl ketones with an example structure that is shown below:



A ketone with a methyl group next to the carbonyl group will show a positive haloform test (yellow precipitate) while aldehydes and non-methyl ketones will not have a reaction in the haloform test.

8. In the table below, fill in the name of the five compounds that were being tested, their structures and their haloform test results.

Compound	Structure	Results of the haloform test
acetone		Yellow precipitate
2-pentanone		Yellow precipitate
3-pentanone		Yellow precipitate
2-hexanone		Positive
2-methyl-2-butanone		Positive

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