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 $2 \text{H}_2 + \text{O}_2 \rightarrow 2 \text{H}_2\text{O}$
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 $2 \text{H}_2 + \text{O}_2 \rightarrow 2 \text{H}_2\text{O}$
 $5 \text{Pb} + \text{OH}^- \rightarrow 2 \text{HCl} + 2 \text{H}_2\text{O} + 1 \text{PbCl}_2$
 $2 \text{AlBr}_3 + 3 \text{K}_2\text{SO}_4 \rightarrow 6 \text{KBr} + 1 \text{Al}_2\text{SO}_4 + 3 \text{CH}_4 + 2 \text{O}_2 \rightarrow 1 \text{CO}_2 + 2 \text{H}_2\text{O}$
 $8 \text{C} + 3 \text{H}_8 + 5 \text{O}_2 \rightarrow 3 \text{CO}_2 + 4 \text{H}_2\text{O}$

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