Chemistry 122 (General College Chemistry I)- Fall 2021 Discussion Worksheet Week 5

Topics: oxidation-reduction reactions, rules for assigning oxidation numbers, oxidizing and reducing agents, stoichiometry and redox titration, combination, decomposition, single displacement and combustion reactions

Relevant Chapters in Silberberg, 9e: 4.4-4.5

Student Directions: Try all problems with your best effort and submit your completed work to Assignments by the due date. Use your text as a resource for any constants or relationships needed which are not given in the problem. Rules for Assigning Oxidation Numbers are found in your text (Table 4.5) on page 174. These must be mastered! NOTE: Some answers are shown in brackets.

You may have to work ahead depending on your discussion day. When topics are fully covered in lecture, go back and redo problems you did not understand. Practice problem solving!

- Give the oxidation number for carbon in each of the following formulas.

 - CF2Cl2 --2.2 = 0 C = +4
 - (c) HCO₂
 - (d) C2H6
 - 7 6 = 0 Z = 5 X = 3 C = 3

 - × + 11 + 0 × + 11 (C + + 11)
- 2. Balance each reaction. Then, classify each reaction as combination, decomposition or displacement. Identify the oxidizing and reducing agents. And answer follow-up questions.
 - (a) $C_2H_2(g) + H_2(g) \rightarrow C_2H_4(g)$

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