# CHEM 120 Exam 2: Chapters 5,6,7,8 (With Answer Key) - Chamberlain College of Nursing

# Examination Study Guide 2023

#### Exam. 2

#### **Total Points: 60**

#### PART 1: Multiple Choice questions: Q 1-24. (2 point each)

- 1. Chemical reactions (equations) show the following
  - a. Reactants.
  - b. Products
  - c. Physical state of each (gas or liquid or solid)
  - d. All of the above
- 2. The following is true for a mole
  - It is also known as Avagadro's number
  - b. It contains 6.022 \* 10<sup>25</sup> molecules
  - c. It contains mass equivalent to molecular weight
  - d. All of the above are true
- 3. In balancing chemical equations, the coefficients are used to
  - a. Show number of moles
  - b. Show volume of each in liters
  - c. Show amount of each in grams
  - d. Show temperature at which the substance is stable
- 4. 2H2 + O2 > 2H2O means that
  - a. 2 grams of H2 g and 1 gram of O2 gas combine to form 2 grams of water.
  - b. 2 moles of H2 g and 1 mole of O2 gas combine to form 2 moles of water.
  - c. 2 grams of H2 g and 1 gram of Q2 gas combine to form 2 moles of water.
  - d. 2 molecules of H2 g and 1 molecule of O2 gas combine to form 2 molecules of water.
- 2H2 + O2 \_\_\_\_\_ > 2H2O, in this equation, correct method to calculate molecular mass of water in
  - a. 2(2)(mass of hydrogen) + mass of oxygen)
  - b. 2(mass of hydrogen gas) + mass of oxygen gas
  - 2)mass of hydrogen) + mass of oxygen

# Pkgacp Chem 120

**Kate Brown** 

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