

CHEM 1110 Equation Sheet Fall 2020

Equations

$$\langle E_k \rangle = \frac{1}{2} m \langle v \rangle^2$$

$$E = \frac{h \cdot c}{\lambda}$$

$$E = h \cdot \nu$$

$$P = k_B(NT/V)$$

$$PV = nRT$$

$$F = \frac{kq_1q_2}{r^2}$$

Units

$$[K] = [^{\circ}C] + 273.15$$

$$1 \text{ atm} = 101,325 \text{ Pa} = 760 \text{ mm Hg}$$

$$1 \text{ cal} = 4.184 \text{ J}$$

Constants

Atomic mass unit	1 amu	=	$1.660538782 \times 10^{-27} \text{ kg}$
	1 g	=	$6.02214179 \times 10^{23} \text{ amu}$
Avogadro's number	N_A	=	$6.02214179 \times 10^{23} / \text{mol}$
Boltzmann's constant	k	=	$1.3806504 \times 10^{-23} \text{ J/K}$
Electron charge	e	=	$1.602176487 \times 10^{-19} \text{ C}$
Faraday's constant	F	=	$9.64853399 \times 10^4 \text{ C/mol}$
Gas constant	R	=	$0.082058205 \text{ L-atm/mol-K}$ $= 8.314472 \text{ J/mol-K}$
Mass of electron	m_e	=	$5.48579909 \times 10^{-4} \text{ amu}$ $= 9.10938215 \times 10^{-31} \text{ kg}$
Mass of neutron	m_n	=	1.008664916 amu $= 1.674927211 \times 10^{-27} \text{ kg}$
Mass of proton	m_p	=	1.007276467 amu $= 1.672621637 \times 10^{-27} \text{ kg}$
Pi	π	=	3.1415927
Planck's constant	h	=	$6.62606896 \times 10^{-34} \text{ J-s}$
Speed of light in vacuum	c	=	$2.99792458 \times 10^8 \text{ m/s}$

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Ying-Ying Zheng



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