

CHEM 1110 Equation Sheet Fall 2020

Equations

$$\langle E_k \rangle = \frac{1}{2} m \langle v \rangle^2$$

$$E = \frac{h \cdot c}{\lambda}$$

$$E = h \cdot \nu$$

$$P = k_B(NT/V)$$

$$PV = nRT$$

$$F = \frac{kq_1q_2}{r^2}$$

Units

$$[K] = [^{\circ}C] + 273.15$$

$$1 \text{ atm} = 101,325 \text{ Pa} = 760 \text{ mm Hg}$$

$$1 \text{ cal} = 4.184 \text{ J}$$

Constants

Atomic mass unit	1 amu	$= 1.660538782 \times 10^{-27} \text{ kg}$
	1 g	$= 6.02214179 \times 10^{23} \text{ amu}$
Avogadro's number	N_A	$= 6.02214179 \times 10^{23} / \text{mol}$
Boltzmann's constant	k	$= 1.3806504 \times 10^{-23} \text{ J/K}$
Electron charge	e	$= 1.602176487 \times 10^{-19} \text{ C}$
Faraday's constant	F	$= 9.64853399 \times 10^4 \text{ C/mol}$
Gas constant	R	$= 0.082058205 \text{ L-atm/mol-K}$ $= 8.314472 \text{ J/mol-K}$
Mass of electron	m_e	$= 5.48579909 \times 10^{-4} \text{ amu}$ $= 9.10938215 \times 10^{-31} \text{ kg}$
Mass of neutron	m_n	$= 1.008664916 \text{ amu}$ $= 1.674927211 \times 10^{-27} \text{ kg}$
Mass of proton	m_p	$= 1.007276467 \text{ amu}$ $= 1.672621637 \times 10^{-27} \text{ kg}$
Pi	π	$= 3.1415927$
Planck's constant	h	$= 6.62606896 \times 10^{-34} \text{ J-s}$
Speed of light in vacuum	c	$= 2.99792458 \times 10^8 \text{ m/s}$

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Table of Contents Pkgacp Chem 1110

1. Understanding the eBook Pkgacp Chem 1110
 - The Rise of Digital Reading Pkgacp Chem 1110
 - Advantages of eBooks Over Traditional Books
2. Identifying Pkgacp Chem 1110
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Pkgacp Chem 1110
 - User-Friendly Interface
4. Exploring eBook Recommendations from Pkgacp Chem 1110
 - Personalized Recommendations
 - Pkgacp Chem 1110 User Reviews and Ratings
 - Pkgacp Chem 1110 and Bestseller Lists

5. Accessing Pkgacp Chem 1110 Free and Paid eBooks
 - Pkgacp Chem 1110 Public Domain eBooks
 - Pkgacp Chem 1110 eBook Subscription Services
 - Pkgacp Chem 1110 Budget-Friendly Options
6. Navigating Pkgacp Chem 1110 eBook Formats
 - ePub, PDF, MOBI, and More
 - Pkgacp Chem 1110 Compatibility with Devices
 - Pkgacp Chem 1110 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Pkgacp Chem 1110
 - Highlighting and Note-Taking Pkgacp Chem 1110
 - Interactive Elements Pkgacp Chem 1110
8. Staying Engaged with Pkgacp Chem 1110
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Pkgacp Chem 1110
9. Balancing eBooks and Physical Books Pkgacp Chem 1110
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pkgacp Chem 1110
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Pkgacp Chem 1110
 - Setting Reading Goals Pkgacp Chem 1110
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Pkgacp Chem 1110
 - Fact-Checking eBook Content of Pkgacp Chem 1110
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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