CHEM 114 Tutorial 1 (Chemical Equilibrium)

 The following reaction represents the binding of oxygen to the protein hemoglobin (Hb), a very important step in respiration and life.

$$Hb(aq) + O2(aq) \Rightarrow Hb:O2(aq)$$

$$\Delta H < 0$$
.

Choose the correct description for how each of the following affects the amount of bound hemoglobin.

- i. Increasing the temperature
 - a. More oxygen will bind and the reaction will shift to the right
 - b. Less oxygen will bind and the reaction will shift to the left
 - There will be no change in the equilibrium
 - d. Not enough information is provided to answer
- ii. Decreasing the concentration of O2 (aq)
 - a. More oxygen will bind and the reaction will shift to the right
 - b. Less oxygen will bind and the reaction will shift to the left
 - c. There will be no change in the equilibrium
 - d. Not enough information is provided to answer
- iii. Decreasing the pressure of O2 (g)
 - a. More oxygen will bind and the reaction will shift to the right
 - b. Less oxygen will bind and the reaction will shift to the left
 - c. There will be no change in the equilibrium
 - d. Not enough information is provided to answer

Pkgacp Chemistry Labs Chem 114

Jicheng Xie

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