

# CHEM 1405 Practice Exam # 3 (Ch 5-7)

Name \_\_\_\_\_

Section # \_\_\_\_\_

**MULTIPLE CHOICE.** Choose the one alternative that best completes the statement or answers the question.

- 1) The ability of an atom to attract the shared electrons in a covalent bond is its \_\_\_\_\_. 1) \_\_\_\_\_  
 A) electronegativity  
 B) nonpolarity  
 C) ionic character  
 D) bonding ability  
 E) polarity
  
- 2) Which of the following substances contains a nonpolar covalent bond? 2) \_\_\_\_\_  
 A) NaCl                      B) N<sub>2</sub>                      C) MgF<sub>2</sub>                      D) NH<sub>3</sub>                      E) H<sub>2</sub>O
  
- 3) A polar covalent bond is found in which of these compounds? 3) \_\_\_\_\_  
 A) H<sub>2</sub>                      B) H<sub>2</sub>O                      C) N<sub>2</sub>                      D) F<sub>2</sub>                      E) NaCl
  
- 4) One mole of copper(II) sulfate, CuSO<sub>4</sub>, contains \_\_\_\_\_ O atoms. 4) \_\_\_\_\_  
 A)  $6.02 \times 10^{23}$   
 B)  $2.41 \times 10^{23}$   
 C) 4  
 D)  $2.41 \times 10^{24}$   
 E)  $1.51 \times 10^{23}$
  
- 5) How many moles of carbon atoms are there in 0.500 mole of C<sub>2</sub>H<sub>6</sub>? 5) \_\_\_\_\_  
 A) 0.500 mole  
 B) 4.00 moles  
 C) 3.00 moles  
 D)  $6.02 \times 10^{23}$  moles  
 E) 1.00 mole
  
- 6) What is the molar mass of sodium phosphate, Na<sub>3</sub>PO<sub>4</sub>? 6) \_\_\_\_\_  
 A) 118.0 g/mole  
 B) 163.9 g/mole  
 C) 226.1 g/mole  
 D) 69.96 g/mole  
 E) 354.0 g/mole
  
- 7) How many moles of water, H<sub>2</sub>O, are present in 75.0 g H<sub>2</sub>O? 7) \_\_\_\_\_  
 A)  $1.35 \times 10^3$  moles  
 B) 75.0 moles  
 C) 4.16 moles  
 D) 7.50 moles  
 E) 4.41 moles

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