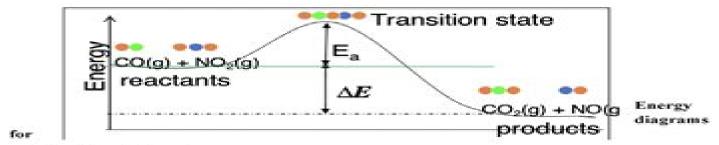
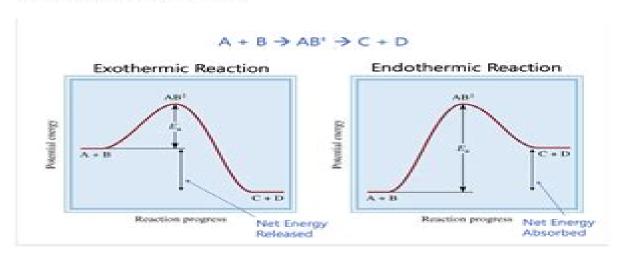
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Energy diagrams:

- Transition state: the high point on the diagram
- Activated complex: the species present at the transition state (temporary and generally
 unstable combination of atoms formed from molecular collisions; will be transformed
 into products as the rxn proceeds
- Activation energy barrier: the energy gap between the reactants and the activated complex
 - Rate depends on the magnitude of the E, required; lower E, means a faster rxn
- Net energy, ΔE: the energy gap between the reactants and the products



exothermic/endothermic rxns:



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