

Chapter 4 Alkanes and Cycloalkanes

IUPAC name

How to name alkane compounds

1. **Identify and name the parent chain** (= the longest consecutive chain of carbons).
 - a. If there is more than one possible parent chain, choose the one with the most substituents attached.
2. **Identify and Name the substituents.**
 - a. Alkane substituents: replace "ANE" with "YL" (e.g. methane → methyl).
 - b. Halogen substituents: replace "INE" with "O" (e.g. fluorine → fluoro).
 - c. Complex substituents
 - i. Number the longest carbon chain **WITHIN** the substituent.
 1. Start with the carbon attached to the parent chain.
 - ii. Name the substituent.
 - iii. Name and Number the substituent's side group.
 - iv. Parenthesis all.
3. **Number the parent chain to assign a locant to each substituent.**
 - a. Chose the direction as the locant # becomes smallest.
 - b. If all #s are same, assign the lowest # alphabetically.
 - c. If there is only one position that can be substituted in the parent chain, do not assign a locant.
 - i. e.g. Chloromethane is not named 1-chloromethane because there is only one position to be substituted.
 - ii. e.g. Mono substituted cycloalkanes do not have numbering because no matter where it's substituted, the locant number is always "1". Chlorocyclohexane is not named 1-chlorocyclohexane.



Chloromethane



Chlorocyclohexane

4. **List the numbered substituents before the parent name in alphabetical order.**
 - a. A prefix is used (di, tri, tetra, penta, etc.) if multiple substituents are identical.
 - i. e.g. Instead of saying "... 3-methyl-5-methyl...", say "... 3,5-dimethyl..."
 - b. Prefix is ignored for alphabetization purposes except for "ISO".
 - c. Hyphen between each substituent; no hyphen between the last substituent and parent name.

Naming cyclic alkanes

1. Monocyclic compounds/substituents.
 - a. Add a prefix "CYCLO" to the total number of carbons consisting the ring system.

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Table of Contents Pkgacp Chem 3411 Lab Manual

1. Understanding the eBook Pkgacp Chem 3411 Lab Manual
 - The Rise of Digital Reading Pkgacp Chem 3411 Lab Manual
 - Advantages of eBooks Over Traditional Books
2. Identifying Pkgacp Chem 3411 Lab Manual
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Pkgacp Chem 3411 Lab Manual
 - User-Friendly Interface
4. Exploring eBook Recommendations from Pkgacp Chem 3411 Lab Manual
 - Personalized Recommendations

- Pkgacp Chem 3411 Lab Manual User Reviews and Ratings
- Pkgacp Chem 3411 Lab Manual and Bestseller Lists
- 5. Accessing Pkgacp Chem 3411 Lab Manual Free and Paid eBooks
 - Pkgacp Chem 3411 Lab Manual Public Domain eBooks
 - Pkgacp Chem 3411 Lab Manual eBook Subscription Services
 - Pkgacp Chem 3411 Lab Manual Budget-Friendly Options
- 6. Navigating Pkgacp Chem 3411 Lab Manual eBook Formats
 - ePub, PDF, MOBI, and More
 - Pkgacp Chem 3411 Lab Manual Compatibility with Devices
 - Pkgacp Chem 3411 Lab Manual Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Pkgacp Chem 3411 Lab Manual
 - Highlighting and Note-Taking Pkgacp Chem 3411 Lab Manual
 - Interactive Elements Pkgacp Chem 3411 Lab Manual
- 8. Staying Engaged with Pkgacp Chem 3411 Lab Manual
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Pkgacp Chem 3411 Lab Manual
- 9. Balancing eBooks and Physical Books Pkgacp Chem 3411 Lab Manual
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pkgacp Chem 3411 Lab Manual
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Pkgacp Chem 3411 Lab Manual
 - Setting Reading Goals Pkgacp Chem 3411 Lab Manual
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Pkgacp Chem 3411 Lab Manual
 - Fact-Checking eBook Content of Pkgacp Chem 3411 Lab Manual

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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