

CHEMISTRY 218 MIDTERM 2022-2023

COMPLETE STUDY GUIDE

independent variable - The experimental factor that is manipulated; the variable whose effect is being studied.

calibration curve - a general method for determining the concentration of a substance in an unknown sample by comparing the unknown to a set of standard samples of known concentration.

How are absorbance and transmittance related? - $A = -\log T$; they are inverses

What are the essential features of a good graph? - labels on the axis, title on the graph, clear data, numbers are well spaced out, and the graph is large enough

Why did you add iron (III) to the aspirin to make your Beer's Law Plot? - Fe^{3+} was added to change the color of the aspirin solution so that the absorbance could be measured in the visible range through spectroscopy.

What is a catalyst and how does it work? - A catalyst is a molecule that is present in the reaction but not consumed; it speeds up a reaction by lowering the activation energy required for the reaction

Explain why a Beer's law plot is linear - Absorbance and concentration must be directly proportional; the higher the concentration of molecules in a solution, the more light that is absorbed and used to excite electrons because there are more particles to absorb the light.

dependent variable - The outcome factor; the variable that may change in response to manipulations of the independent variable.

linear scale - the same distance along the axis always represents the same increase in the variable

interpolation - estimating a value within the range of measured data

extrapolation - projection; conjecture; V. extrapolate: infer (unknown information) from known information

lambda max - the wavelength that gives the highest absorbance value for a sample

What physical process is being observed using spectroscopic measurements? - The excitation of electrons through the absorption of wavelengths

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Boyles



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