

MODULE 1

GUIDE QUESTION

The electron is represented as a particle in fixed orbits around the nucleus in the Bohr Model. The electron is mathematically described as a wave in the Quantum Mechanical Model. To explain the distribution of electrons in the atom, three coordinates, or three quantum numbers, were necessary.

MODULE 2

1. Al: $1s^2 2s^2 2p^4 3s^2 3p^3$ - $1s^2 2s^2 2p^6 3s^2 3p^1$: the mistake is $2p^4$ and the correct is $2p^6$
2. B: $1s^2 2s^2 2p^5$ - $1s^2 2s^2 2p^1$: the mistake is $2p^5$ and the correct is $2p^6$
3. F: $1s^2 2s^2 2p^6$ - $1s^2 2s^2 2p^5$: the mistake is $2p^6$ and the correct is $2p^5$

The configuration that violates Pauli exclusion principle and Hund's rule is letter d because it leaves the box for $2s$ blank. Letter e and b also violates the principle and rule because one in the orbital diagram is pointing upward instead of downward.

MODULE 3

If I had to choose which type of bond I want to make with other individuals, I'd go with a covalent bond. Because sharing is one of the keys to a strong link, if one of the atoms needs something to share with the other, the other atom is always there to get or gain it. I only need one atom who I can always count on to share or show my good and bad electrons and accept me for who I am, and who will always be there no matter what happens. I also chose covalent bond because of the triple bond form, which depicts the three elements necessary for a successful partnership: trust, loyalty, and honesty. It also symbolizes the three most significant aspects of each atoms' lives: god, family, and the future. As a result,

MODULE 4

Atoms involved	Type of bond	Lewis dot symbol of each atom	Charge of each ion after electron transfer if ionic bond is formed	Lewis dot symbol of each ion if ionic bond is formed	Formula of the product
Na, Cl	Ionic	Na Cl	Na Cl	Na Cl	NaCl
C, I	Covalent	C I	Not applicable	Not applicable	CI ₄
Mg, Cl	Ionic	Mg Cl	Mg Cl	Mg Cl	MgCl ₂
Ca, F	Ionic	Ca F	Ca F	Ca F	CaF ₂
Na, O	Ionic	Na O	Na O	Na O	Na ₂ O
Ca, N	Ionic	Ca N	Ca N	Ca N	Ca ₂ N ₂
S, Cl	Covalent	S Cl	Not applicable	Not applicable	SCl ₂ q

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