

CHEM-111 Elements of General Chemistry

Fall 2017 – Sections 501 & 502

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Office Hours: Tue 1:00 – 2:45 pm, Wed 1:30 – 4:30 pm,
Thu 1:00 – 2:45, and anytime by appointment

Meeting Times: Mon & Wed (501) Lecture: 12:00 – 1:15 pm, VAAS 127
Mon (501) Lab/Recitation: 1:30 – 4:15 pm, VAAS 128
Tue & Thu (502) Lecture: 3:00 – 4:15 pm, VAAS 127
Tue (502) Lab/Recitation: 4:30 – 7:15 pm, VAAS 128

COURSE DESCRIPTION

One-semester course in general chemistry, especially for non-science majors in the health sciences except pre-medicine and medical technology. (Credit not allowed for both CHEM 111 and CHEM 121L.) Meets New Mexico Lower Division General Education Common Core Curriculum Area III: Science (NMCCN 1114). Prerequisite: ACT => 22 or SAT => 510 or MATH 103 or MATH 121 or MATH 150 or MATH 162 or MATH 163 or MATH 180 or MATH 181 or MATH 264.

WHAT YOU'LL LEARN

COURSE TEACHING & LEARNING OUTCOMES

By the end of this course, student will be able to:

1. Use dimensional analysis, the SI system of units and appropriate significant figures to express quantities, convert units and perform quantitative calculations in science.
2. Diagram the structure of the atom in terms of its subatomic particles; and justify the existence and nature of the subatomic particles and the scale of the nucleus using appropriate experiments from scientific history.
3. Use the IUPAC system of nomenclature and knowledge of reaction types to describe chemical changes, predict products and represent the process as a balanced equation.
4. Apply the mole concept to amounts on a macroscopic and microscopic level and use this to perform stoichiometric calculations including for reactions in solution and gases.
5. Apply the gas laws and kinetic molecular theory to relate atomic level behavior to macroscopic properties.
6. Describe the ways in which atoms combine to form molecules (ionic and covalent). Apply knowledge of electronic structure to determine molecular structure, geometry and hybridization.

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