

3. Do you expect propanol to be miscible with water? Why or why not?

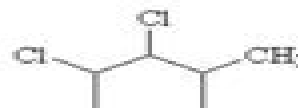
Miscible.  $C_3H_7OH$  is small and has dipole-dipole and hydrogen bonds like  $H_2O$

4. Which of the following chemical formulas could be a cycloalkane?

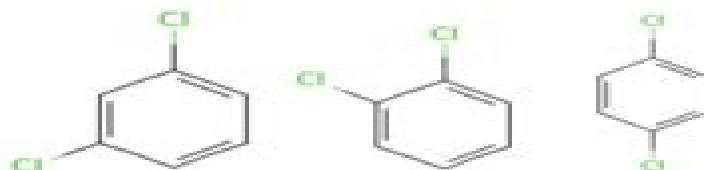
- a.  $C_5H_{12}$
- b.  $C_2H_4$
- c.  $C_6H_{10}$
- d.  $C_4H_{10}$

5. What is the IUPAC name for this molecule?

- a. 1-methyl-2,3-dichlorocyclopentane
- b. 1,2-dichloro-3-methylcyclopentane
- c. 1,2-dichloro-3-methylcyclobut-1-ane
- d. 3-methyl-1,2-dichlorocyclopent-1-ane
- e. 1,2-dichloro-3-methyl-1-pentane



6. Name the following molecules. Which, if any, are nonpolar?



*meta*-dichlorobenzene, *ortho*-dichlorobenzene, *para*-dichlorobenzene

The only nonpolar one is *para*-dichlorobenzene because the Cl's cancel out. These molecules are planar.

7. According to VSEPR, what geometry does ethyne have?

- a. Linear
- b. Bent
- c. Trigonal Planar
- d. Tetrahedral

# Pkg Acp Chem 130

**BM King**



**Pkg Acp Chem 130:**

This book delves into Pkg Acp Chem 130. Pkg Acp Chem 130 is a vital topic that needs to be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Pkg Acp Chem 130, encompassing both the fundamentals and more intricate discussions.

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