

LAB #3: ACID-BASE TITRATION

DATA & CALCULATIONS

PART A: Titration of Sulphuric Acid Solution (the unknown)

TABLE #1: Titration of 25.00 mL of H_2SO_4 acid solution with NaOH solution

Trial	1	2	3	4	5
Initial Buret Volume (mL)	1.35	1.41	1.00	0.85	1.11
Final Buret Volume (mL)	26.18	26.50	26.61	27.17	26.73
Volume of NaOH (mL)	24.83	25.09	25.61	26.32	25.62

PART B: Standardization of NaOH Solution (secondary standard) with Potassium Phthalate (primary standard)

Mass of empty beaker _____ 50.550 _____ g

Mass of beaker + potassium hydrogen phthalate _____ 55.500 _____ g

Mass of potassium hydrogen phthalate _____ 4.950 _____ g

TABLE #2: Titration of 25.00 mL of KHP with NaOH solution

Trial	1	2	3	4	5
Initial Buret Volume (mL)	1.50	1.35	0.89	1.08	0.90
Final Buret Volume (mL)	26.72	27.75	26.97	27.39	27.40
Volume of NaOH used (mL)	25.22	26.40	26.08	26.31	26.50

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