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Quiz #4 – Combustion Analysis Name: Shaili Batsri

CHEM 1411 – Please give Empirical & Molecular Formula!

1) A 0.450 g sample of an oily liquid substance containing C, H & O produces 1.023 g CO₂ and 0.419 g H₂O in a combustion analysis. What is the empirical formula?

Empirical Formula: C₃H₆O

2) The oily liquid smells like pineapple, is a key ingredient used as a flavor enhancer in processed orange juices, and has a molar mass is approximately 116 g/mol, what is the molecular formula?

Molecular Formula: C₆H₁₂O₂

3) Write a balanced chemical equation for the above substance.



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1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Pkg Acp Kingwood C Chem 1411
- Chapter 2: Essential Elements of Pkg Acp Kingwood C Chem 1411
- Chapter 3: Pkg Acp Kingwood C Chem 1411 in Everyday Life
- Chapter 4: Pkg Acp Kingwood C Chem 1411 in Specific Contexts
- Chapter 5: Conclusion

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4. In chapter 3, the author will examine the practical applications of Pkg Acp Kingwood C Chem 1411 in daily life. This chapter will showcase real-world examples of how Pkg Acp Kingwood C Chem 1411 can be effectively utilized in everyday scenarios.

5. In chapter 4, this book will scrutinize the relevance of Pkg Acp Kingwood C Chem 1411 in specific contexts. This chapter will explore how Pkg Acp Kingwood C Chem 1411 is applied in specialized fields, such as education, business, and technology.

6. In chapter 5, this book will draw a conclusion about Pkg Acp Kingwood C Chem 1411. This chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Pkg Acp Kingwood C Chem 1411.

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