

CHEM-116-2515 - Intro Organic & Biochemistry

Experiment Activity 3 Tests of Aldehydes and Ketones Key

Aldehydes and ketones both contain the carbonyl group, $C=O$. Their functional groups have very similar structure. In this experiment, we are going to learn more about aldehydes and ketones by looking at their reactions and perform tests on a variety of aldehydes and ketones to distinguish between them.

Watch this video <https://youtu.be/gdzziTrFsE4> up to 11:45 minutes only (skip aldol condensation reaction) and answer the following questions.

1. Tollen's silver mirror test is used to distinguish between aldehydes and ketones. In the experiment, five compounds were tested with the Tollen's reagent. Name and draw the five compounds.



2. In the Tollen's silver mirror test, a decanting technique was used. What is decanting?
Decanting is carefully pouring out the clear liquid while leaving behind the unwanted solids.
3. Which of the five compounds produced a positive Tollen's silver mirror test (formation of a silver mirror)?
Formalin and acetyl aldehyde instantaneously formed a silver color.
Benzaldehyde should have reacted but in this experiment, the reaction was too slow to occur, so no reaction was observed.
4. Which of the five compounds should produce a positive Tollen's silver mirror test but the reaction was too slow to be observed?
Benzaldehyde
5. Which of the five compounds produced a negative Tollen's silver mirror test? Acetone and 3-pentanone
6. What can you conclude from the result of the Tollen's silver mirror test? Provide an explanation.
Conclusion: aldehydes show a positive Tollen's silver mirror test while ketones show a negative Tollen's silver mirror test.

Explanation: Tollen's reagent is an oxidizing agent. Aldehydes can be oxidized by the Tollen's reagent show a positive result. Ketones cannot be oxidized so they show a negative Tollen's silver mirror test.

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