

SP2023 CHEM 214
General Chemistry Laboratory II

Experiment 3: Synthesis of coordination compound and
Experiment 4: Redox titration of the oxalate ligand in the prepared coordination
compound

Pre and Post Laboratory Questionnaire
Laboratory Report [Combined]

Name of the Student:

Student ID:

Date of Submission:

Guidelines:

- Submission must be made through the assignment option created on **BB**
- You are required to complete the given file and submit it as a soft copy [handwritten/typed]

Pre-Lab Questions:

1. 1) How many grams of Ammonia (NH₃) can be produced from the reaction of 28.0 grams of Nitrogen (N₂) and 25.0 grams of Hydrogen (H₂)?



a) Known:

28.0 g N₂ and 25.0 g H₂

b) Unknown: ? g NH₃

c) Formula Weights:

FW. of N₂ = 28.014 g N₂ = 1 mol N₂

FW. of H₂ = 2.0158 g H₂ = 1 mol H₂

FW. of NH₃ = 17.0307 g NH₃ = 1 mol NH₃

d) Mole Relationships:

1 mol N₂ = 2 mol NH₃

3 mol H₂ = 2 mol NH₃

e) Solutions:

Limiting

Reagent:

Excess

Reagent:

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C Cleary



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