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web an solution initially contains a catalytic amount of an enzyme with $k_m = 1.5 \text{ mmol/L}$ of substrate and no product after 45 seconds the solution contains $25 \text{ }\mu\text{mol}$ of product find v_{max} and the concentration of product after 20 minutes hint $s = k_m$ *reaction mechanisms practice kinetics khan academy* - Jan 30 2023

web a reaction and its experimental rate law are represented below $2 \text{ ICl(g)} + \text{H}_2 \text{ (g)} \rightarrow 2 \text{ HCl(g)}$ rate $k [\text{ICl}]^2 [\text{H}_2]$ a chemist proposes a mechanism for the reaction that is consistent with the rate law the mechanism has two elementary steps and the first step is slow compared to the second

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consider the following mechanism
 $\text{O}_3 \xrightarrow{\text{fast}} \text{O}_2 + \text{O}$
 $\text{O} + \text{O}_2 \xrightarrow{\text{slow}} \text{O}_3$
 a) write the overall balanced chemical equation
 b) identify any intermediates within the mechanism

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 $\text{Mg} + 2\text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$
 the reaction between Mg and HCl is represented by the equation above in a kinetics experiment
 a) 0.080 g sample of Mg ribbon is added to an Erlenmeyer flask containing 50 mL of 3.0 M HCl at 25 °C

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 reactions conducted in an industrial facility mix compounds together heating and stirring them for a

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when he arrived he was told not to work in the mine by leah but he refused leah harboured him when he got to the mine he
was made the head mine boy he directed the affairs of the boys and even the red one respected his opinion

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