

Arba Minch University
Department of Chemistry
School of post Graduate
Studies

Non-aqueous Solution
Chemistry

Feb. 2024

Non-aqueous Solution Chemistry

- ✓ Many inorganic reactions take place in aqueous solution,
 - ✓ **water is not always a suitable solvent; some reagents may react with H_2O (e.g. the alkali metals) and non-polar molecules are insoluble in water.**
- ✓ The use of solvents other than water is commonplace for the organic chemist, and such compounds include **dichloromethane, hexane, toluene** and **ethers** such as **diethyl ether, tetrahydrofuran, and diglyme**
- ✓ non-aqueous solvents categories into :
 - * **protic solvents** (e.g. HF , H_2SO_4 , $MeOH$);
 - * **aprotic solvents** (e.g. N_2O_4 , BrF_3);
 - * **coordinating solvents** (e.g. $MeCN$, Et_2O , Me_2CO).

Non Aqueous Solution Chemistry

Ying-Ying Zheng



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