

Lost At Sea New Edition

SJ Ball



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web by joseph le fanu buy study guide carmilla summary the novella is set in 19th century styria which is located in austria it

is narrated by a young woman named laura who tells her story to doctor hesselius whose papers are being organized by an aide laura lives with her father a widower in a castle deep in the forest of styria

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web carmilla s mother is described as an elegant well dressed older woman she is very mysterious and is the one who introduces carmilla to her victims and then leaves in the cases of both bertha and laura carmilla s mother leaves her daughter in the care of these young ladies households

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web historical context of carmilla is set in austria during the late 1800s and le fanu clearly draws upon contemporary issues of the victorian era this was a time that was noteworthy for its social and industrial reform and for the formation and rise of

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web carmilla also known as mircalla and millarca is the story s eponymous antagonist a vampire from an old aristocratic family she appears eternally as a beautiful young woman and preys on vulnerable young women to whom she is sexually attracted

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web elements in the second row of the periodic table place their electrons in the 2n shell as well as the 1n shell for instance lithium li has three electrons two fill the 1 s orbital and the third is placed in the 2 s orbital giving an electron configuration of 1 s 2 2 s 1

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web one more point needs to be emphasized about the relationship between electron configuration and the periodic table the atoms of elements in the same vertical column of the table have similar electron configurations for example consider the alkaline earth elements group iia

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web the electron removed during the ionization of beryllium he 2s 2 is an s electron whereas the electron removed during the ionization of boron he 2s 2 2p 1 is a p electron this results in a lower first ionization energy for boron even though its nuclear charge is greater by one proton

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web predictions from reliable sources have been used for these elements grayed out electron numbers indicate subshells filled to their maximum bracketed noble gas symbols on the left represent inner configurations that are the same in each period written out these are he 2 helium 1s 2 ne 10 neon 1s 2 2s 2 2p 6

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web q1 what is the electron configuration of the element potassium answer the atomic number of potassium is 19 a neutral potassium element has 19 electrons its electron configuration is 1s 2 2s 2 2p 6 3s 2 3p 6 4s 1 or simply put ar 4s 1 q2 what is the total number of electrons that can be filled in s p and d subshells

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