



Shadow Over The Caribbean

Sasha Hu



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therefore the strongest acid is $\text{C}_2\text{H}_3\text{COOH}$ q2 lime water is a CaOH_2 c CaCO_3 d CaCl_2 correct answer b CaOH_2

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$\text{aq} \text{C}_3\text{H}_7\text{NO}_2 + \text{aq} \text{H}_3\text{O}^+ \rightleftharpoons \text{aq} \text{C}_3\text{H}_8\text{NO}^+ + \text{aq} \text{H}_2\text{O}$ l $\text{CH}_3\text{O}_2\text{H} + \text{aq} \text{NH}_3 \rightleftharpoons \text{aq} \text{CH}_3\text{CO}_2^- + \text{aq} \text{NH}_4^+$ aq $\text{SbF}_5 + \text{aq} \text{HF} \rightleftharpoons \text{aq} \text{SbF}_6^- + \text{aq}$ identify the

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$\text{CaCO}_3 + 2\text{HBr} \rightarrow \text{CaBr}_2 + \text{H}_2\text{O} + \text{CO}_2$

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each other acids mixed with bases neutralize each other acids mixed with bases make stronger bases acids mixed with bases

make stronger acids acids and bases interact with each other in what is called a neutralization reaction

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solution a solution of a weak acid in water is a mixture of the nonionized acid hydronium ion and the conjugate base of the

acid with the nonionized acid present in the greatest concentration

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