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Lauren Gardner



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 $2 \text{ CO}_2(\text{g}) + \text{H}_2\text{O}(\text{g}) + 2 \text{ C}_2\text{H}_2(\text{g}) + 5 \text{ O}_2(\text{g}) \rightarrow 2 \text{ C}_2\text{H}_6(\text{g}) + 4 \text{ H}_2\text{O}(\text{g})$
 $\Delta h = 94.5 \text{ kJ}$
 $\text{H}_2(\text{g}) + \frac{1}{2} \text{ O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{g})$
 $\Delta h = 71.2 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g}) + \frac{7}{2} \text{ O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 3 \text{ H}_2\text{O}(\text{g})$
 $\Delta h = -1560 \text{ kJ}$

co₂ g 3 h 2 o g Δh 283 kj g 1 c₂h₆ g

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