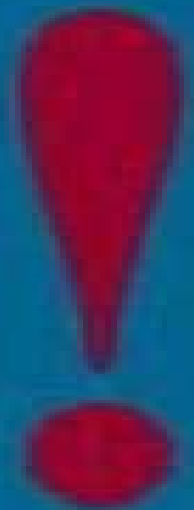


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web each gas in a mixture of gases exerts its own pressure independently of the other gases present therefore the pressure of each gas within a mixture is called the partial pressure of the gas dalton s law of partial pressures states that the total pressure of a mixture of gases is equal to the sum of the partial pressures of the component gases

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following questions about ideal gas law 41 what is the formula

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web at very high pressures a gas will occupy a larger volume than predicted by the ideal gas law which of the following statements is true at low temperatures intermolecular forces become important and the pressure of a gas will be lower than predicted by

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