

CHEM 114 Tutorial 1 (Chemical Equilibrium)

1. The following reaction represents the binding of oxygen to the protein hemoglobin (Hb), a very important step in respiration and life.



Choose the correct description for how each of the following affects the amount of bound hemoglobin.

i. Increasing the temperature

- a. More oxygen will bind and the reaction will shift to the right
- b. Less oxygen will bind and the reaction will shift to the left
- c. There will be no change in the equilibrium
- d. Not enough information is provided to answer

ii. Decreasing the concentration of O₂ (aq)

- a. More oxygen will bind and the reaction will shift to the right
- b. Less oxygen will bind and the reaction will shift to the left
- c. There will be no change in the equilibrium
- d. Not enough information is provided to answer

iii. Decreasing the pressure of O₂ (g)

- a. More oxygen will bind and the reaction will shift to the right
- b. Less oxygen will bind and the reaction will shift to the left
- c. There will be no change in the equilibrium
- d. Not enough information is provided to answer

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