

max. Correctly 30 Points

Student's name: Andriana ID: 000000

2

5

These 10 questions are to assess your knowledge of the questions. Questions without corresponding answers will receive 0 points. Answers without proper notes will have partial credit deducted. This option is subject to your choice.

1. A compound is a substance composed of:

- ☐ a) molecules
- ☒ b) an ionic substance
- ☐ c) metal
- ☐ d) something that forms a solid
- ☐ e) at least two elements

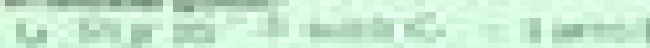
2. The equation that describes the transport of water across a membrane is called the:

- 3
- ☐ a) osmotic pressure
 - ☒ b) molecular formula
 - ☐ c) law of conservation of mass
 - ☐ d) transport formula
 - ☐ e) law of conservation of energy

3. Consider CH_4 is one of the most common compounds in nature. Write the complete formula for methane.



4. Methane (CH_4) is commonly used as a fuel in combustion reactions. What is the empirical formula of methane gas?



5. A gas has a mass of 1.0 g and a volume of 1.0 L. The gas is composed of 1.0 g of carbon and 1.0 g of hydrogen. What is the empirical formula of the gas?

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